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Supplementary Information for

Disproportionate increase in freshwater methane emissions induced by experimental warming

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This Supplementary Information includes:

Supplementary Discussion

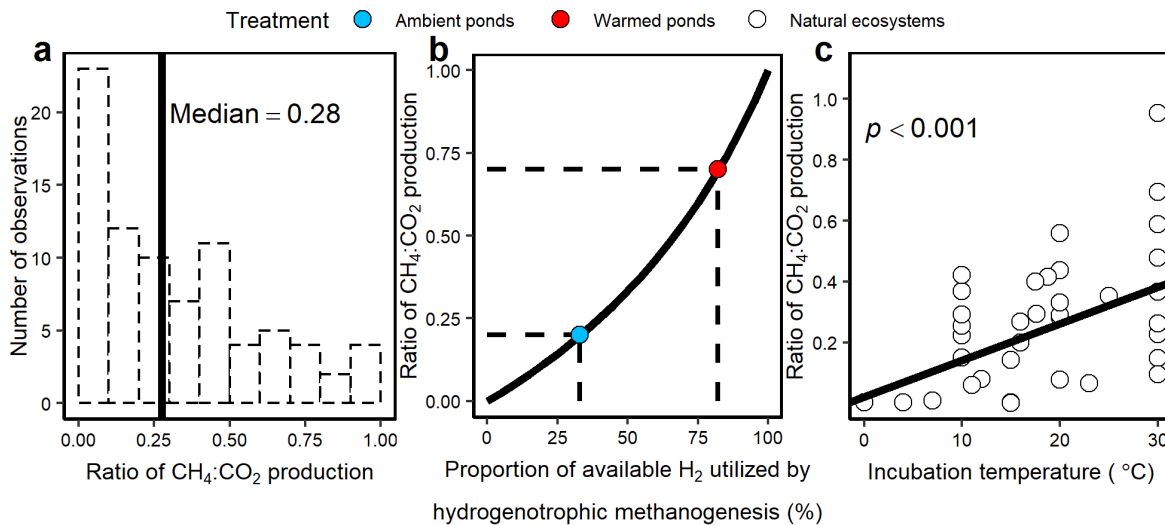
Supplementary Fig. 1 to 6

Supplementary Tables 1 to 10

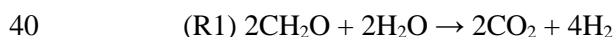
22 Supplementary Discussion

23 The production of both CH₄ and CO₂ by a combination of acetoclastic and hydrogenotrophic 24 methanogenesis in relation to Ralf Conrad's 1999 publication (Ref. 24 in main text)

25 Determining the absolute ratio of CH₄:CO₂ from any substrate via methanogenesis is challenging. Conrad
26 (Ref. 24 in main text) calculated the idealised outcome of glucose degradation in a strictly methanogenic
27 system, which does return a 1:1 ratio of CH₄ and CO₂, via 33% hydrogenotrophic and 67% acetoclastic
28 methanogenesis. However, he then shows that this idealised ratio is rarely true in nature where a strictly
29 methanogenic system simply does not exist (and *see* Ref. 28 and 34 cited in the main text). Indeed here,
30 we have compiled CH₄ and CO₂ production data from 13 studies including wetlands¹⁻¹¹, permafrost
31 thaw¹² and lakes¹³ which demonstrate that the vast majority of CH₄:CO₂ production ratios are less than 0.5
32 with a median of 0.28 only (*see* panel **a** in the figure below). Thus, in reality, Conrad's idealised ratio
33 appears to be rare in natural systems and the same is true in our experimental ponds where the CH₄:CO₂
34 ratios are less than 1:1 in both the ambient and warmed ponds (0.2:1 vs. 0.7:1).

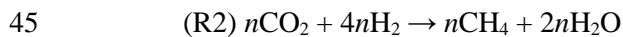


35
36 Given that the carbon quality is similar between the warmed and ambient ponds (*see* Fig. 2b in main text),
37 why has the CH₄:CO₂ ratio increased after eleven years' warming? Here Conrad's model can be used to
38 consider what a change in the CH₄ to CO₂ ratio might mean. Conrad's concept (ref. 24) through
39 fermentation assumes that:

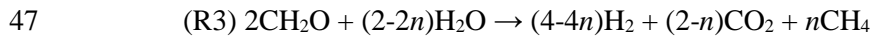


41 In most freshwater ecosystems where inorganic electron acceptors other than CO₂ are not available,
42 hydrogenotrophic methanogenesis competes against homoacetogenesis for electrons from H₂. We can

43 assume that the proportion of available H₂ utilized by hydrogenotrophic methanogenesis is n ($0 < n < 1$) and
44 by multiplying everything in hydrogenotrophic methanogenesis with n as a factor we get:



46 The sum of reaction (R1) and (R2) is:



48 The ratio of CH₄ to CO₂ produced is therefore $n/(2-n)$. The 1:1 ratio of CH₄ to CO₂ production occurs
49 only when 100% of H₂ produced via fermentation (R1) is used up to reduce CO₂ to CH₄ (i.e., $n=1$), the
50 CH₄:CO₂ production ratio is however $< 1:1$ when homoacetogenesis outcompetes hydrogenotrophic
51 methanogenesis for H₂ and electrons flow to acetate rather than CH₄ ($n < 1$). More importantly, (R3)
52 predicts that the ratio of CH₄ to CO₂ increases exponentially as a function of the proportion of available
53 H₂ being utilized by hydrogenotrophic methanogenesis (n) (*see panel b* in the figure above). As H₂
54 becomes more available at higher temperatures¹³, should an increasing CH₄:CO₂ ratio with temperature be
55 expected? Indeed, this hypothesis is validated by a positive correlation between incubation temperatures
56 and CH₄:CO₂ ratios produced in anoxic wetland soils^{1,6,8,14} (*see panel c* in the figure above). Therefore, an
57 idealized 1:1 ratio is rare in reality but the CH₄:CO₂ ratio increases towards the idealized 1:1 ratio
58 predicted by Conrad at higher temperatures.

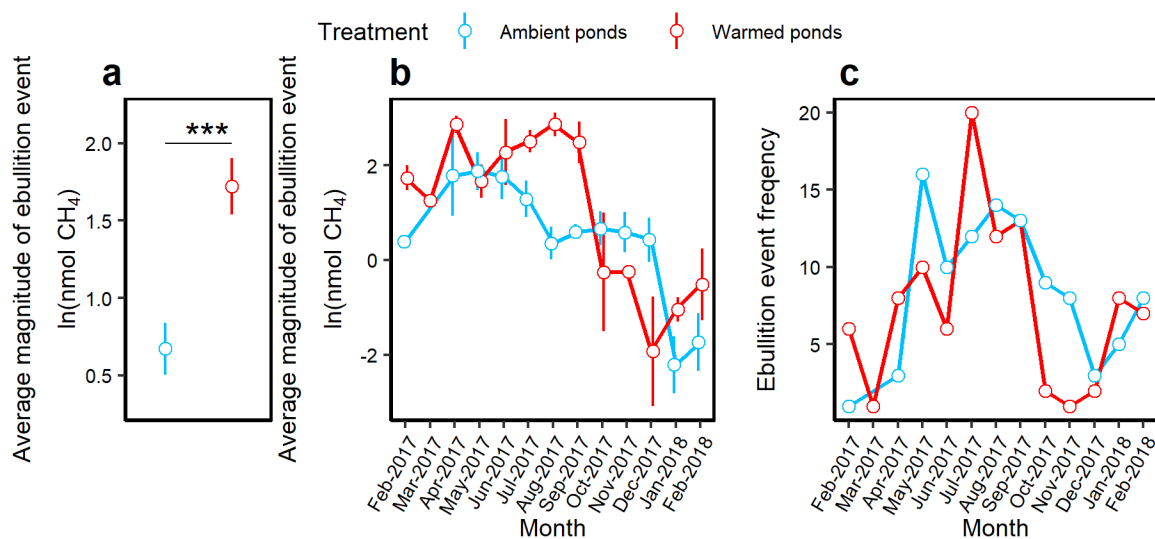
59 Furthermore, now we can rationalize the disproportionate increase in CH₄:CO₂ production ratio seen in
60 our long-term warmed ponds using the proportion of available H₂ being utilized by hydrogenotrophic
61 methanogenesis. At lower temperatures, electrons flow to acetate and as a result methane production is
62 dominated by acetoclasty¹⁵ and only a minor proportion of available H₂ is utilized by hydrogenotrophic
63 methanogenesis (30 %, the blue dot in panel **b**, *see* figure above). In contrast, as H₂ concentrations
64 increase with temperature, which thermodynamically favours hydrogenotrophic methanogenesis, a larger
65 proportion of electrons and carbon flow to CH₄ (80 %, the red dot in panel **b**, *see* figure above), ultimately
66 increasing the CH₄ to CO₂ ratio closer towards the idealised ratio predicted by Conrad. In Figure 3d of the
67 main text, we show clearly that hydrogenotrophic methanogenesis is favoured by warming and so
68 conclude that, from whatever source, more of the available hydrogen is being directed more efficiently
69 into methane in the warmed compared to the ambient ponds and that such disproportionate increase in
70 CH₄:CO₂ ratio will probably occur in natural freshwaters as the Earth warms.

71

72 **Supplementary Figures**

73 **Supplementary Fig. 1 | Magnitude and frequency of methane emission through ebullition events**
 74 **($n=198$, 1.2% identified of the 16504 total chamber measurements using our two criteria) and**
 75 **exclusion of 7 other non-steady flux events – see Methods).**

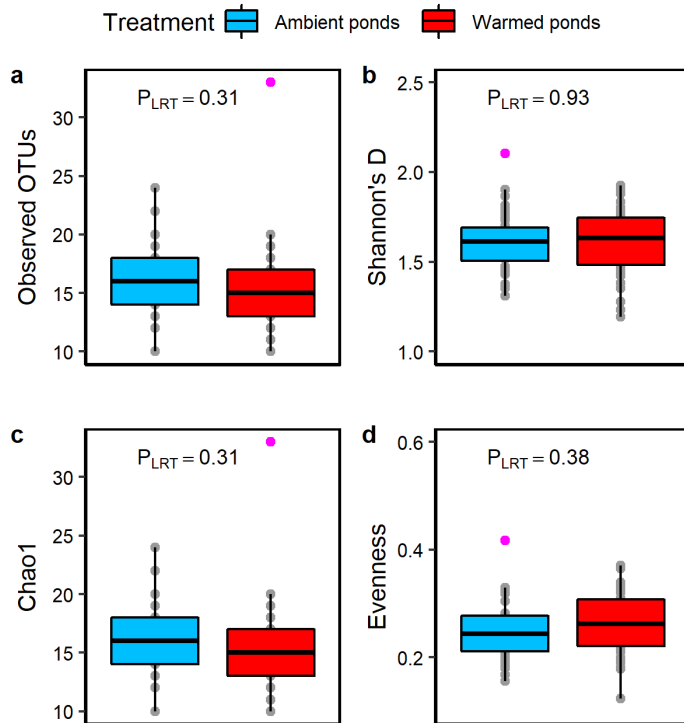
76 Ebullition in our ponds exports methane from the sediments to the atmosphere directly and therefore: **1**,
 77 should increase with enhanced methanogenesis under warming; and **2**, follow a similar seasonal pattern to
 78 diffusion. Indeed, in line with our enhanced methanogenesis under warming, the average magnitude of
 79 ebullition events **a**, was 3-fold greater in the warmed ponds (80 ng CH₄ per event versus 27 ng CH₄ per
 80 event in the ambient ponds, *t*-statistic, ***: $p < 0.001$). In addition, the magnitude of ebullition events **b**,
 81 and their frequency **c**, peaked in summer, demonstrating a similar seasonal pattern to diffusional methane
 82 emissions (Extended Data Fig. 2). Ebullition events in our ponds have therefore been captured. However,
 83 ebullition contributed only 0.2% of total methane emissions in both warmed and ambient ponds. The
 84 magnitude of ebullition events was calculated using the maximum methane concentration in a chamber
 85 measurement - see equation (1) in Methods. Error bars are standard errors of the magnitude of an
 86 ebullition event.



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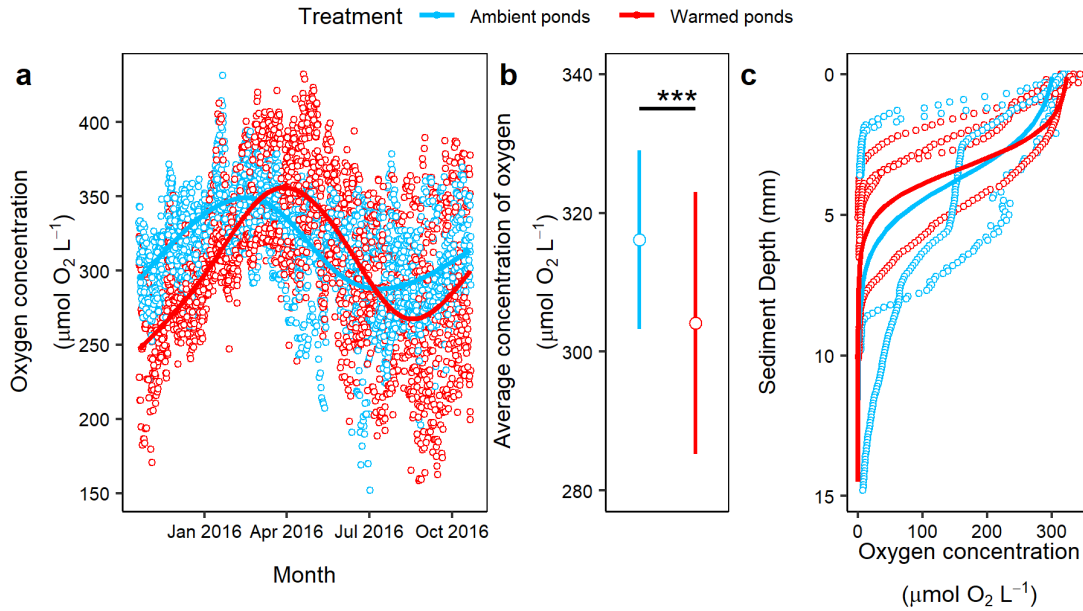
89 **Supplementary Fig. 2 | Methanogen alpha diversity ($n=79$, monthly samples from April to August**
 90 **in 2016 from 8 ambient and 8 warmed ponds, see Methods). a**, Observed OTUs, **b**, Shannon's
 91 diversity, **c**, Chao 1 diversity and **d**, evenness are all identical between the warmed (red) and ambient
 92 (blue) ponds. Statistical significance (P_{LRT}) was determined by a likelihood ratio test. Box lower and
 93 upper bounds are 25th and 75th percentiles, respectively, the line is the median. Whiskers indicate
 94 largest/smallest value no further than 1.5X the interquartile range. The data points (in magenta) beyond
 95 the end of whiskers are outliers.



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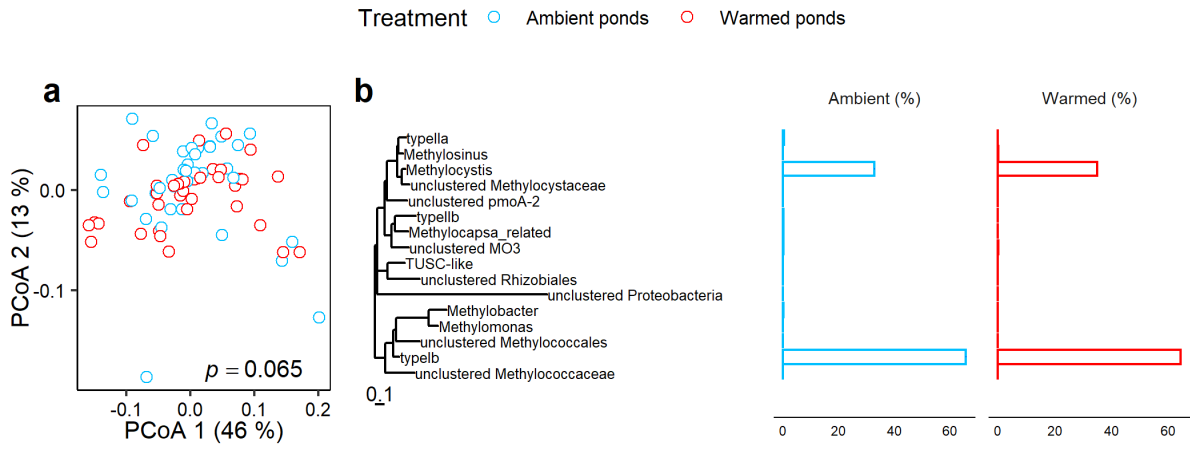
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98 **Supplementary Fig. 3 | Water column and sediment oxygen concentrations in the experimental**
99 **ponds. a**, Seasonality of the *in situ* dissolved oxygen concentrations in the overlying water of the warmed
100 (red) and ambient (blue) ponds from October, 2015 to October, 2016 ($n=5120$, data collected at 10-
101 minute intervals using oxygen sensor in 7 ambient and 7 warmed). **b**, Mean *in situ* dissolved oxygen
102 concentration was lower in the warmed ponds compared to their ambient controls ($n=5120$, t -statistic,
103 $p<0.001$). **c**, Oxygen penetration profiles measured in intact sediment cores at 15 °C ($n=6$, from 3 warmed
104 and 3 ambient in April, 2016). Oxygen concentrations showed a steeper decline and penetrated to a
105 shallower depth in the warmed pond sediment (4.86 mm) compared to 6.67 mm to the ambient pond
106 sediments.



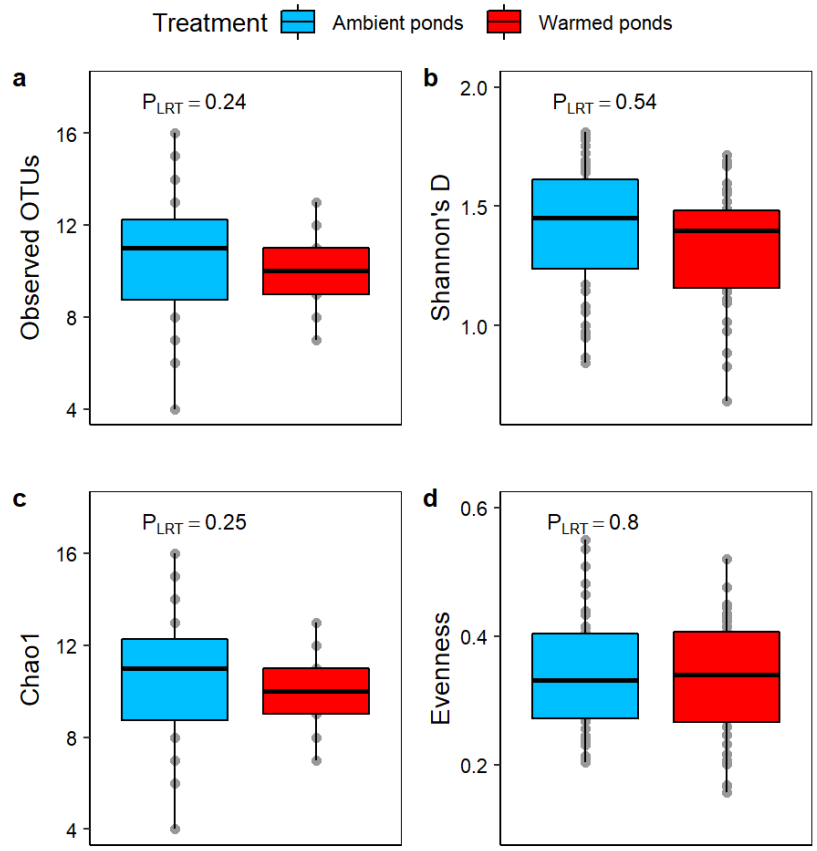
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109 **Supplementary Fig. 4 | Effect of long-term warming on the methanotroph community composition**
 110 **($n=80$, monthly samples from March to July in 2017 from 8 ambient and 8 warmed ponds, see**
 111 **Methods). a**, No overall change in the methanotroph community with long-term warming demonstrated
 112 by principal coordinate analysis (PCoA) using Bray-Curtis analysis and a Hellinger standardized dataset
 113 (at genus level) and **b**, differential abundance analysis at genus level detected no significant changes in
 114 any methanotroph genus.



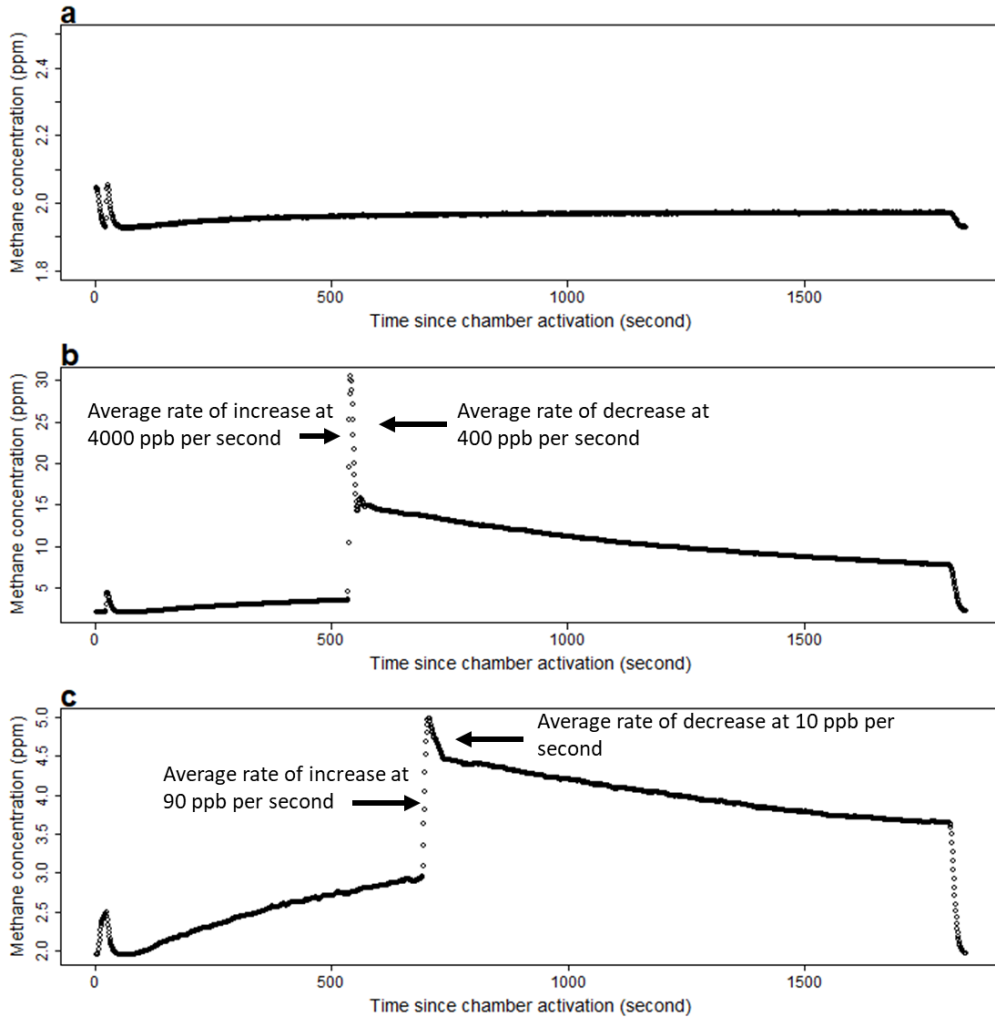
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117 **Supplementary Fig. 5 | Methanotroph alpha diversity ($n=80$, monthly samples from March to July**
 118 **in 2017 from 8 ambient and 8 warmed ponds, see Methods). a, Observed OTUs, b, Shannon's**
 119 **diversity, c, Chao 1 diversity and d, evenness are identical between the warmed (red) and ambient (blue)**
 120 **ponds. Statistical significance (P_{LRT}) was determined by a likelihood ratio test. Box lower and upper**
 121 **bounds are 25th and 75th percentiles, respectively, the line is the median. Whiskers indicate**
 122 **largest/smallest value no further than 1.5X the interquartile range.**



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125 **Supplementary Fig. 6 | Example of chamber measurements for a, steady-state flux b, strong**
126 **ebullition and c, gentle ebullition.** Chambers with a steady state flux (a) had standing methane
127 concentrations of ~2 ppm. When a strong ebullition event occurred methane rose very rapidly to 30 ppm
128 at ~ 4,000 ppb/s, while, in gentler ebullition events (c), methane concentrations could increase at 90 ppb/s
129 to ~5 ppm. In both cases methane concentrations subsequently decreased more gently than the rapid
130 increase.



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Supplementary Tables

135 **Supplementary Table 1 | Original data sources for the analysis of methane emission capacities from**
 136 **globally distributed sites.** Lat and Long represent latitude and longitude of each site. Avg. Temp =
 137 average annual temperature in each site. *n* represents the number of daily rate measurements of methane
 138 emissions for each site. The *p* values are less than 0.05 for each site, representing a good relationship
 139 between methane emission and air-temperature.

Site ID	Site Name	Lat	Long	Avg. Temp (°C)	Type	<i>n</i>	<i>p</i> value	
AT-Neu	Neustift	47.12	11.31	10.0	Grassland	539	<0.05	
CA-SCB	Scotty Creek Bog	61.31	-121.3	11.8	Wetland	639	<0.05	16
FR-LGt	La Gvette	47.32	2.28	13.4	Wetland	215	<0.05	
US-CRT	Curtice Walter-Berger cropland	41.63	-83.35	7.2	Cropland	246	<0.05	17
US-EML	Eight Mile Lake Permafrost thaw gradient, Healy Alaska.	63.88	-149.25	3.6	Open shrubs	1015	<0.05	18
US-LA1	Pointe-aux-Chenes Brackish Marsh	29.50	-90.45	22.9	Wetland	206	<0.05	19
US-LA2	Salvador WMA Freshwater Marsh	29.86	-90.29	24.0	Wetland	531	<0.05	20
US-Los	Lost Creek	46.08	-89.98	6.9	Wetland	1499	<0.05	21
US-Myb	Mayberry Wetland	38.05	-121.77	17.1	Wetland	2687	<0.05	22
US-ORv	Olentangy River Wetland Research Park	40.02	-83.02	13.3	Wetland	1132	<0.05	23
US-OWC	Old Woman Creek	41.38	-82.51	20.3	Wetland	104	<0.05	24
US-PFa	Park Falls/WLEF	45.95	-90.27	6.9	Forest	975	<0.05	25
US-Sne	Sherman Island Restored Wetland	38.04	-121.76	16.1	Wetland	575	<0.05	26
US-StJ	St Jones Reserve	39.09	-75.44	18.0	Wetland	250	<0.05	27
US-Tw1	Twitchell West Pond Wetland	38.11	-121.65	18.6	Wetland	2039	<0.05	28
US-Tw4	Twitchell East End Wetland	38.10	-121.64	17.5	Wetland	1668	<0.05	29
US-Twt	Twitchell Island	38.11	-121.65	18.2	Cropland	351	<0.05	30
US-Uaf	University of Alaska, Fairbanks	64.87	-147.86	12.0	Forest	236	<0.05	31
US-WPT	Winous Point North Marsh	41.46	-82.99	11.3	Wetland	793	<0.05	32

140 **Supplementary Table 2 | Annual methane budget, pond water characteristics and pond sediment**
 141 **characteristics.**

		Ambient	Warmed	Ratio (W/A)
Production	Methane production capacity at 15 °C ¹ (MG _{T15} , μmol CH ₄ m ⁻² d ⁻¹)	2795 (1092)	7086 (2767)	2.5
	Effect of 4 °C warming predicted using the apparent activation energy $\overline{E_{MP}}$ (Effect _{warming}) ²	1	1.5	1.5
	Methane production capacity (totMG)³	2795	10274	3.7
	<i>mcrA</i> abundance (log ₁₀ (copy g ⁻¹ (wet sediments)))	6.59 (0.045)	6.77 (0.034)	1.5
	Methanogen cell-specific activity (fmol CH ₄ <i>mcrA</i> ⁻¹ h ⁻¹)	0.35	0.59	1.7
Emission and proportion of CH ₄ oxidized <i>in situ</i>	Annual methane emission (ME, μmol CH ₄ m ⁻² d ⁻¹)	233 (22)	562 (63)	2.4
	Amount of methane oxidized <i>in situ</i>⁴ (<i>in situ</i> totMO, μmol CH₄ m⁻² d⁻¹)	2563	9713	3.8
	Proportion of methane oxidized <i>in situ</i> ⁵ (MO%, %)	92	95	1.03
	Required proportion of CH₄ oxidized (%_{pred})⁶		98	
Oxidation	Kinetic effect of <i>in situ</i> methane concentrations (Effect _{kinetic}) ⁷	1	1.9	1.9
	Effect of 4 °C warming predicted using apparent activation energy $\overline{E_{MO}}$ (Effect _{warming}) ²	1	1.4	1.4
	Effect of sampling depth (Effect _{sampling}) ⁸	1	1.4	1.4
	Methane oxidation capacity (<i>ex situ</i> totMO)⁹			3.6
	<i>pmoA</i> abundance (log ₁₀ (copy g ⁻¹ (wet sediments)))	3.99 (0.047)	4.38 (0.038)	2.45
	Predicted fold increase in <i>pmoA</i> abundance to offset warming-induced methane production (Ab _{pred}) ¹⁰			2.67
	Methanotroph cell-specific activity (pmol CH ₄ <i>pmoA</i> ⁻¹ h ⁻¹)	25.0	10.2	0.4
Water characteristic	Dissolved CH ₄ concentration (μmol L ⁻¹)	0.51 (0.15)	1.07 (0.21)	2.1
Sediment characteristic	Sediment % carbon	0.83 (0.089)	1.23 (0.13)	1.48
	Sediment % nitrogen	0.084 (0.0061)	0.11 (0.0010)	1.31
	Sediment C:N	9.37 (0.41)	10.40 (0.31)	1.11

142 Numbers given in the brackets are standard errors.

143 1. Methane production capacity at 15 °C was calculated by taking the exponential of the ln-transformed

144 methane production rate in equation (4) ($\overline{\ln F(T_C)}$) and converting from nmol g⁻¹ h⁻¹ to μmol m⁻² d⁻¹

145 (sediment density 1,068 Kg m⁻³ and depth 0.08 m) (MG_{T15} = $\overline{\ln F(T_C)} \times 10^6 \times 24 \times 1.068 \times 10^6 \times 0.08$)

146 2. Effect of 4 °C warming on methane production and oxidation was calculated from the apparent
147 activation energies: \overline{E}_{MP} and \overline{E}_{MO} for methane production and oxidation, respectively (*see* equations (2)
148 and (4)). Apparent activation energies for methane production and oxidation in the warmed ponds are 0.7
149 eV and 0.57 eV, respectively, predicting a 1.5- and 1.4-fold increase in the methane production and
150 oxidation, respectively.

151 3. In total, methane production capacity in the warmed ponds increased by 3.7-fold
152 ($\text{totMG} = \text{MG}_{T15} \times \text{Effect}_{\text{warming}}$).

153 4. Total methane oxidized *in situ* is the difference between methane production capacity and annual
154 methane emission (i.e., *in situ* totMO = totMG - ME)

155 5. Proportion of oxidized methane is the percentage of methane emission to methane production capacity
156 at annual average temperatures (i.e., $\text{MO}\% = (\text{totMG} - \text{ME}) / \text{totMG} \times 100\%$).

157 6. Proportion of methane oxidation required in the warmed ponds to prevent methane emissions from
158 increasing (i.e., $(\text{totMG}_{\text{warmed}} - \text{ME}_{\text{ambient}}) / \text{totMG}_{\text{warmed}} \times 100\%$).

159 7. Methane oxidation capacities at *in situ* methane concentrations were calculated using Michaelis-
160 Menten model based on the methane concentrations in the pond water (*see* equation (7)).

161 8. Oxygen penetrated 4.86 and 6.67 mm into the warmed and ambient pond sediments, respectively
162 (Supplementary Fig. 3). These depths were used as proxy for the active methanotrophy layer. Therefore,
163 the effect of sampling the same depths in the warmed and ambient ponds for methane oxidation capacity
164 measurements is $\text{Effect}_{\text{sampling}} = \frac{20 \text{ mm}}{4.86 \text{ mm}} / \frac{20 \text{ mm}}{6.67 \text{ mm}}$.

165 9. In total, warming increased the measured methane oxidation capacity in the warmed ponds by 3.6-fold
166 ($\text{Effect}_{\text{kinetic}} \times \text{Effect}_{\text{warming}} \times \text{Effect}_{\text{sampling}}$), accounting for the discrepancy between predicted and measured
167 methane emissions *in situ* (*ex situ* totMG = *in situ* totMO).

168 10. Warming has increased the methane oxidation capacity by 3.6-fold but not the 3.9-fold required to
169 offset the greater warming-induced methane production (i.e., $(\text{totMG}_{\text{warmed}} - \text{ME}_{\text{ambient}}) / (\text{totMG}_{\text{ambient}} -$
170 $\text{ME}_{\text{ambient}})$). Predicted methanotroph abundance (Ab_{pred}) to offset the greater warming-induced methane
171 production is therefore the abundance of methanotroph required to achieve the predicted 3.9-fold methane
172 oxidation capacity in warmed pond sediment if the efficiency per methanotroph stays the same (i.e., 10.2
173 $\text{pmol CH}_4 \text{ pmoA}^{-1} \text{h}^{-1}$).

174 *In situ* methane oxidation is limited by the diffusion of methane and oxygen. We acknowledge that by
175 mixing the sediments and ^{13}C -CH₄ in our laboratory slurry measurements we would have optimized the

176 methane oxidation capacity in the sediments from both the warmed and ambient ponds. Therefore, we
177 represent here, for the *ex situ* methane oxidation capacity, only the ratio between the warmed and ambient
178 pond sediments (ratio W/A) to show that the kinetic effect and temperature effect increased the methane
179 oxidation capacity by 1.9- and 1.4-fold in the warmed ponds relative to their ambient counterparts,
180 respectively. In addition, if the depth of oxygen penetration serves as a proxy for active methanotrophy
181 layer, altogether, the *ex situ* methane oxidation capacity in the warmed ponds would be 3.6-fold higher
182 than in the ambient controls, close to the increase in methane production in the warmed ponds i.e. 3.7-
183 fold, as well as the predicted amount of methane oxidized *in situ* i.e. 3.8-fold.

184

185 **Supplementary Table 3 | Linear mixed-effect model results for the β -diversity analysis of *mcrA***
 186 **library ($n=79$, monthly samples from April to August in 2016 from 8 ambient and 8 warmed ponds,**
 187 ***see Methods*) and *pmoA* library ($n=80$, monthly samples from March to July in 2017 from 8 ambient**
 188 **and 8 warmed ponds, *see Methods*).**

189 The β -diversity was estimated using the scores along the first two principle coordinate axis (PCoA1 and
 190 PCoA2) of the Bray-Curtis distance measures. These were then fitted into a mixed-effects model with
 191 pond and sampling month treated as random effects. The statistical significance of the treatment (i.e.,
 192 ambient or warmed ponds) was determined from the *F*-test using Satterthwaite's method for denominator
 193 degrees-of-freedom and *F*-statistic. The results were similar to a PERMANOVA analysis.

Treatment	PCoA1			PCoA2		
	% Variation ¹	F-value	P-value	% Variation ¹	F-value	P-value
<i>mcrA</i>	34.04	6.13	<0.05	22.93	3.54	<0.10
<i>pmoA</i>	46.20	0.037	0.85	13.89	0.46	0.51

194 ¹. The amount of variation captured in the axis is defined as the proportion of eigenvalue of that axis
 195 to the sum of all eigenvalues.

196
 197

198 **Supplementary Table 4 | Taxonomy assignment to the *mcrA* OTUs at 85 % identity ($n=79$, monthly**
 199 **samples from April to August in 2016 from 8 ambient and 8 warmed ponds, see Methods). Numbers**
 200 **in parentheses are standard errors.**

Family	Genus	Sequence reads	
		Ambient	Warmed
<u>Hydrogenotrophic methanogens</u>			
unclustered <i>Methanomicrobiales</i>	unclustered <i>Methanomicrobiales</i>	227,766 (558)	225,753 (1,304)
<i>Methanospirillaceae</i>	<i>Methanospirillum</i>	256,777 (583)	184,696 (1,169)
<i>Methanobacteriaceae</i>	<i>Methanobacterium</i>	69,265 (210)	107,752 (398)
<i>Methanomicrobiaceae</i>	<i>Methanoplanus</i>	316 (3)	320 (7)
<i>Methanomicrobiaceae</i>	<i>Methanomicrobium</i>	174 (2)	223 (5)
<i>Methanocellaceae</i>	<i>Methanocella</i>	149 (3)	83 (2)
<i>Methanothermaceae</i>	<i>Methanothermus</i>	141 (1)	161 (2)
<i>Methanocaldococcaceae</i>	<i>Methanocaldococcus</i>	69 (1)	13 (0.3)
<i>Methanobacteriaceae</i>	<i>Methanothermobacter</i>	0 (0)	31 (1)
<i>Methanomicrobiaceae</i>	<i>Methanoculleus</i>	0 (0)	73 (2)
<u>Acetoclastic methanogens</u>			
<i>Methanosaetaceae</i>	<i>Methanosaeta</i>	257,132 (606)	287,992 (1,345)
<i>Methanosarcinaceae</i>	unclustered <i>Methanosarcinaceae</i>	7,285 (42)	5,071 (44)
<u>Methylotrophic methanogens</u>			
unclustered <i>Thermoplasmata</i>	unclustered <i>Thermoplasmata</i>	648 (7)	271 (7)
<i>Methanosarcinaceae</i>	<i>Methanohalophilus</i>	394 (4)	1,012 (8)
<i>Methanoplasmatales</i>	unclustered <i>Methanoplasmatales</i>	46 (1)	28 (1)
<i>Methanosarcinaceae</i>	<i>Methanosalsum</i>	0 (0)	25 (1)
<i>Methanosarcinaceae</i>	<i>Methanomethylovorans</i>	0 (0)	320 (8)

201

202 **Supplementary Table 5 | Taxonomy assignment to the *pmoA* OTUs at 90 % identity (*n*=80, monthly**
 203 **samples from March to July in 2017 from 8 ambient and 8 warmed ponds, see Methods).**
 204 Highlighted warmed pond data were at a lower relative abundance in those ponds compared to the
 205 ambient ponds. Numbers in parentheses are standard errors.

Family	Genus	Sequence reads	
		Ambient	Warmed
<i>Methylococcaceae</i>	Type Ib	713,570 (1,689)	597,859 (1,081)
<i>Methylocystaceae</i>	<i>Methylocystis</i>	357,045 (1,166)	324,967 (855)
<i>Methylocystaceae</i>	Type IIa	4,784 (59)	100 (2)
MO3	unclustered MO3	2,532 (53)	1,681 (26)
<i>Beijerinckiaceae</i>	<i>Methylocapsa</i> -related	2,399 (37)	5 (0.1)
<i>Environmental samples</i>	Type IIb	2,012 (26)	79 (1)
<i>Methylococcaceae</i>	TUSC-like	1,062 (12)	14 (0.4)
<i>Methylococcaceae</i>	<i>Methylobacter</i>	1,372 (13)	158 (3)
<i>Methylocystaceae</i>	<i>Methylosinus</i>	535 (7)	1,848 (18)
unclustered <i>Proteobacteria</i>	unclustered <i>Proteobacteria</i>	362 (5)	0 (0)
pmoA-2	unclustered pmoA-2	235 (4)	410 (5)
<i>Methylocystaceae</i>	unclustered <i>Methylocystaceae</i>	166 (2)	88 (2)
unclustered <i>Rhizobiales</i>	unclustered <i>Rhizobiales</i>	149 (2)	12 (0.3)
<i>Methylococcaceae</i>	<i>Methylomonas</i>	13 (0.3)	0 (0)
unclustered <i>Methylococcales</i>	unclustered <i>Methylococcales</i>	0 (0)	181 (4)
<i>Methylococcaceae</i>	unclustered <i>Methylococcaceae</i>	0 (0)	28 (1)

206
207

208 **Supplementary Table 6 | Multi-model selection for fitting generalized additive mixed effects models**
 209 **to the seasonal CH₄ emission data (*n*=3553 steady-state estimates, representing emissions from 7**
 210 **ambient and 7 warmed ponds with each pond being notionally measured three times per day – see**
 211 **Methods).**

212 To assess the effect of long-term warming on the median rate of methane emissions, a range of
 213 generalized additive mixed effects models (GAMMs) were fitted to the daily methane emission rate data
 214 (ME) as a function of treatment (i.e., warmed or ambient pond) and day of the year since 1st January 2017
 215 (DOY). Whether the seasonal pattern of methane emissions differed between the treatment was also
 216 tested by comparing the smoother terms (DOY, by=Treatment) and s(DOY). Models were ranked using
 217 the AIC. ΔAIC refers to differences in AIC relative to the smallest AIC value and AIC weight is the
 218 probability of any model providing the best fit to the data e.g., 0.913 indicates that model (1) is the best fit
 219 to the data.

	Model	d.f.	AIC	ΔAIC	AIC Weight
(1)	Ln(ME)~Treatment+s(DOY,by=Treatment)	8	12035.0	0.00	0.913
(2)	Ln(ME)~s(DOY,by=Treatment)	7	12039.7	4.71	0.087
(3)	Ln(ME)~Treatment+s(DOY)	6	12194.2	159.24	0.000
(4)	Ln(ME)~s(DOY)	5	12198.9	163.94	0.000

220 A GAMM which included treatment on the intercept and a treatment-specified smoother term for DOY
 221 provided the best fit to the seasonal methane emission data, demonstrating an increase in median methane
 222 emission from warmed ponds as well as a difference in seasonality.
 223
 224

225 **Supplementary Table 7 | Multi-model selection for fitting linear mixed-effect models to CH₄**
 226 **potential production data (nmol CH₄ g⁻¹h⁻¹) as a function of treatment (e.g. warmed or ambient**
 227 **ponds), additional substrates and experimental incubation temperature (Ts) (n=662, sediment**
 228 **samples collected monthly and randomly from 3 to 5 of the 10 ambient and 3 to 5 of the 10 warmed**
 229 **ponds with each incubated at 3 temperatures and with up to 2 substrates. The sample size for**
 230 **control only, i.e., without additional substrates, of the total 662 samples, was 238. No replicate was**
 231 **applied within each pond.).**

232 Ts represents the standardized temperature $\left(\frac{1}{kT_c} - \frac{1}{kT_{ij}}\right)$ in equation (4). A range of linear mixed-effect
 233 models were fitted to the rate of methane production (ln(MG)) data. Note that only the fixed-effect parts
 234 of the models are included in the table. As in Supplementary Table 6, models were ranked using the AIC
 235 and an AIC weight of 0.76 indicates that model 1 is the best fit to the data.

	Model	d.f.	AIC	ΔAIC	AIC Weight
	Ln(MG)~Ts+Treatment+Substrate				
(1)	+Ts×Treatment+Ts×Substrate +Treatment×Substrate	13	1820.6	0.00	0.76
	Ln(MG)~Ts+Treatment+Substrate				
(2)	+Ts×Treatment+Ts×Substrate+ Treatment×Substrate+Ts×Treatment×Substrate	15	1823.4	2.78	0.19
	Ln(MG)~Ts+Treatment+Substrate				
(3)	+Ts×Substrate+Treatment×Substrate	12	1826.3	5.71	0.04
	Ln(MG)~Ts+Treatment+Substrate				
(4)	+Ts×Treatment+Ts×Substrate	11	1829.2	862	0.01
	Ln(MG)~Ts+Treatment+Substrate+Ts×Substrate				
(5)		10	1834.5	13.94	0.001

236 Note that in the incubations above 22°C the rate of CH₄ production plateaued and therefore we excluded
 237 these data from the model.
 238

239 **Supplementary Table 8 | Model selection procedure for fitting linear mixed-effect models to**
 240 **sediment methane potential production data (MG) as a function of carbon turnover k ($n=32$,**
 241 **sediment samples collected from 4 ambient and 4 warmed ponds in April, May, June and August,**
 242 **2017).**

243 The full model included additive terms and their interactions for two fixed effects – natural logarithm of
 244 carbon turnover k ($\ln k$) and treatment type (i.e., ambient or warmed ponds). The significance (p -values) of
 245 the fixed-effect terms was determined using a likelihood ratio test on nested models. The p -value for
 246 comparing “Treatment $\times\ln k$ ” was 0.40, and the term removed from the model. As removing
 247 “Treatment $\times\ln k$ ” had no significant effect on model fit, model F1, that included a single slope but distinct
 248 intercepts provided the best fit to the methane potential data (marked in bold), demonstrating that the
 249 potential of sediments to produce methane increased equally in both the warmed and ambient pond
 250 sediments as carbon quality also increased but warming has stepped-up the fraction of carbon turned to
 251 methane.

Model	d.f.	AIC	LogLik	p -value
F0) $\ln(\text{MG}) \sim \ln k \times \text{Treatment} + \ln k + \text{Treatment}$	6	94.2	-41.09	
F1) $\ln(\text{MG}) \sim \ln k + \text{Treatment}$	5	92.9	-41.44	0.40
F2) $\ln(\text{MG}) \sim \ln k$	4	98.8	-45.4	<0.01
F3) $\ln(\text{MG}) \sim 1$	3	107.7	-50.9	<0.001

252
 253

254 **Supplementary Table 9 | Model selection procedure for fitting mixed-effect models to methane**
 255 **oxidation as a function of kinetic or temperature responses.**

256 **a, Fitting Michaelis-Menten models to CH₄ oxidation rate (MO) as a function of initial CH₄**
 257 **concentration ($n=158$, sediment samples collected from 8 ambient and 8 warmed ponds in July,**
 258 **2017, and December, 2018, with a range of initial methane concentrations, see equation (7) in**
 259 **Methods).** The full model included the initial methane concentrations (C_{CH_4}) and the two Michaelis-
 260 Menten parameters defining the kinetic response, i.e., the maximum methane oxidation rate (V_{max}) and the
 261 Michaelis constant (K_m). The significance of treatment (i.e., warmed or ambient) on the parameters (V_{max}
 262 + Treatment) and (K_m + Treatment) was determined via Likelihood Ratio Test on nested models. As
 263 removing the effect of treatment on K_m and V_{max} had no significant effect on model fit ($p=0.98$ and
 264 $p=0.45$, respectively), the model with the same V_{max} and K_m terms for both warmed and ambient pond
 265 sediments provided the best fit to the CH₄ oxidation data (model F2, marked on bold).

Model	d.f.	AIC	LogLik	p -value
F0) MO ~ (C_{CH_4} , V_{max} +Treatment, K_m +Treatment)	6	2106.21	-1047.10	
F1) MO ~ (C_{CH_4} , V_{max} +Treatment, K_m)	5	2104.21	-1047.10	0.98
F2) MO ~ (C_{CH_4}, V_{max}, K_m)	4	2102.78	-1047.39	0.45

266

267 **b, Model selection procedure for fitting linear mixed-effect models to the temperature sensitivity of**
 268 **CH₄ oxidation rate (ln(MO)) ($n=192$, sediment samples collected from 8 warmed and 8 ambient**
 269 **ponds in May, June and July, 2017, incubated under four different temperatures).** The full model
 270 included additive terms and their interactions for two fixed effects – standardized temperature at 15 °C
 271 (Ts, term $\left(\frac{1}{kT_c} - \frac{1}{kT_{ij}}\right)$ in equation (4)) and treatment type (i.e., ambient or warmed ponds). The
 272 significance of fixed-effect terms (p -values) were determined using a likelihood ratio test on nested
 273 models. For example, the significance of the term “Treatment×Ts”, i.e., distinct slopes between warmed
 274 and ambient pond sediments, was determined by comparing nested model F0 to its reduced model F1.
 275 The p -value of this comparison was 0.24, the term “Treatment×Ts” was not significant and was thus
 276 removed from the model. As removing “Treatment×Ts” or “Treatment” had no significant effect on
 277 model fit, the model F2, that included a single slope and intercept, therefore provided the best fit to the
 278 CH₄ oxidation rate data (marked in bold), demonstrating that the CH₄ oxidation capacity and its
 279 temperature sensitivity were the same in both the warmed and ambient ponds.

Model	d.f.	AIC	LogLik	χ^2	p -value
F0) ln(MO)~Ts+Treatment×Ts+Treatment	8	287.12	-135.56		
F1) ln(MO)~Ts+Treatment	7	286.48	-135.56	1.36	0.24
F2) ln(MO)~Ts	6	287.81	-137.91	3.33	0.068
F3) ln(MO)~1	5	329.75	-159.88	43.94	<0.001

280

281 **Supplementary Table 10 | Model selection procedure for fitting linear mixed-effect models to the**
 282 **carbon conversion efficiency (CCE) data.**

283 **a, Carbon conversion efficiency as function of temperature** ($n=191$, sediment samples collected from
 284 8 warmed and 8 ambient ponds in May, June and July, 2017, incubated under four different
 285 temperatures): The full model included additive terms and their interactions for two fixed effects –
 286 centered temperature at 15 °C (T_c , term $T - T_c$ in equation (8)) and treatment types (i.e., ambient or
 287 warmed ponds). The significance (p -values) of fixed-effect terms was determined using a likelihood ratio
 288 test on nested models. As removing “Treatment \times T_c ” or “Treatment” had no significant effect on model
 289 fit, the model, F2, that included one common slope and intercept provided the best fit to CCE data
 290 (marked in bold).

Model	d.f.	AIC	LogLik	χ^2	p -value
F0) CCE~ T_c +Treatment+ Treatment \times T_c	8	1041.2	-512.61		
F1) CCE~ T_c +Treatment	7	1039.5	-512.72	0.23	0.63
F2) CCE~T_c	6	1037.7	-512.84	0.23	0.63
F3) CCE~1	5	1069.0	-529.51	33.34	<0.01

291
 292 **b, Carbon conversion efficiency as function of methane concentration** ($n=69$, sediment samples
 293 collected in July, 2017, from 8 ambient and 8 warmed ponds with a range of initial methane
 294 concentrations): The full model included additive terms and their interactions for two fixed effects –
 295 initial ^{13}C - CH_4 concentration (C_{CH_4} , *see* equation (9)) and treatment type (i.e., ambient or warmed ponds).
 296 The significance (p -value) of fixed-effect terms was determined using a likelihood ratio test on nested
 297 models. As removing “ $C_{\text{CH}_4}\times$ Treatment” or “Treatment” had no significant effect on model fit, the model,
 298 F2, that included one common slope and intercept provided the best fit to CCE data (marked in bold).

Model	d.f.	AIC	LogLik	χ^2	p -value
F0) CCE~ C_{CH_4} +Treatment+ $C_{\text{CH}_4}\times$ Treatment	6	458.78	-223.39		
F1) CCE~ C_{CH_4} +Treatment	5	456.86	-223.43	0.083	0.77
F2) CCE~C_{CH_4}	4	455.63	-223.82	0.77	0.38
F3) CCE~1	3	463.93	-228.97	10.30	<0.01

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