**Article**

**Isothiourea-Catalyzed Enantioselective α-Alkylation of Esters via 1,6-Conjugate Addition to para-Quinone Methides**

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**Abstract:** The isothiourea-catalyzed enantioselective 1,6-conjugate addition of para-nitrophenol esters to 2,6-disubstituted para-quinone methides is reported. para-Nitrophenoxide, generated in situ from initial N-acylation of the isothiourea by the para-nitrophenol ester, is proposed to facilitate catalyst turnover in this transformation. A range of para-nitrophenol ester products can be isolated, or derivatized in situ by addition of benzylamine to give amides at up to 99% yield. Although low diastereorecontrol is observed, the diastereoisomeric ester products are separable and formed with high enantiocontrol (up to 94:6 er).

**Keywords:** isothiourea; ammonium enolate; arylxide; quinone methide; ester functionalization; 1,6-conjugate addition

1. Introduction

Quinone methides (QMs) are electrophilic compounds composed of a cyclohexadiene core bearing a carbonyl either ortho or para to an exocyclic alkylidene unit [1,2]. Due to their electrophilicity [3–6], QMs have been used in a variety of biological and medicinal processes [1,2,7], are present within natural products and pharmaceuticals [1,2,8–11], and have been applied as electrophiles in a variety of synthetic reactions [1,2,12–18]. While ortho-QMs have been used extensively in enantioselective catalysis [19], particularly as components in formal [4+2] cycloaddition reactions, the use of para-QMs has only recently received increased attention [19–23]. The majority of enantioselective organocatalytic methods that involve para-QMs have utilized Bronsted acid [24–33] or hydrogen bonding catalysts [34–38], with only a relatively small number of examples using Lewis base catalysis [39–51].

C(1)-Ammonium enolate intermediates [52–55], generated by the reaction of a tertiary amine Lewis base catalyst with a ketene, anhydride or acyl imidazole [56], have found widespread application for the synthesis of heterocyclic scaffolds in high yield and with excellent enantiocontrol. Traditionally these approaches have been limited by the requirement for the electrophilic reaction partner to contain a latent nucleophilic site to facilitate catalyst turnover. This conceptual obstacle has resulted in catalysis via C(1)-ammonium enolates being mostly applied for formal cycloaddition reactions. More recently, aryl esters have emerged as alternative C(1)-ammonium enolate precursors [55,57,58]. Significantly, following acylation of the tertiary amine catalyst by the aryl ester, a nucleophilic arylxide is liberated, which may be exploited again in the catalytic cycle to facilitate ammonium enolate formation and catalyst turnover (Scheme 1a) [55,59]. This strategy offers a potentially...
general solution to allow the expansion of electrophile scope within catalytic processes using C(1)-ammonium enolate intermediates. In 2014, we applied this concept for the isothiourea-catalyzed[2,3]-rearrangement of allylic ammonium ylides (Scheme 1b) [60–64]. More recently, this approach has been used by Snaddon (Scheme 1c) [65–72], Hartwig (Scheme 1d) [73] and Gong [74,75] for co-operative isothiourea/transition metal-catalyzed α-functionalization of pentafluorophenyl esters. In both cases, an isothiourea-derived C(1)-ammonium enolate is intercepted by an electrophilic transition metal complex to affect an allylation or benzylation reaction. We have expanded the scope of electrophiles applicable within this catalyst turnover strategy to include iminium ions generated under either photoredox conditions or Brønsted acid catalysis, as well as bis-sulfone Michael acceptors, and pyridinium salts (Scheme 1e) [76–79]. Recently, Waser also reported an elegant example of this turnover strategy for the enantioselective α-chlorination of pentafluorophenyl esters [80], while Zheng and co-workers reported a significant challenge within this area is the identification of electrophilic reaction partners that react with the catalytically-generated C(1)-ammonium enolate, but are compatible with the nucleophilic tertiary amine catalyst and aryloxide, which is essential for catalyst turnover. Building upon this conceptual platform, it was envisaged that para-QMs may be suitable electrophiles to apply in formal 1,6-conjugate additions.

**Scheme 1.** Isothiourea-catalyzed enantioselective processes using aryloxide-facilitated catalyst turnover.
2. Results

2.1. Reaction Optimization

Initial studies focused on the isothiourea-catalyzed 1,6-conjugate addition of para-nitrophenyl (PNP) ester 1 to 2,6-di-tert-butyl para-QM 5 (Table 1). Benzylamine was added at the end of the reaction to convert the PNP ester product to the corresponding amide. Based on our previous experience, the amide was expected to be more stable to chromatographic purification. Using tetramisole HCl 6 (20 mol%) as the catalyst, i-Pr₂NEt as the base and MeCN as the solvent gave a 55:45 ratio of diastereoisomeric amide products 7 and 8 in high yield (82%) and with excellent enantioselectivity (7: 97:3 er; 8: 94:6 er) (Entry 1). Chromatographic separation of the diastereoisomers was not possible, however, the enantioenrichment of both 7 and 8 could be reliably determined by chiral stationary phase (CSP)-HPLC analysis of the mixture. A control reaction in the absence of the catalyst showed no conversion (Entry 2). The use of six alternative solvents was investigated (PhMe, CH₂Cl₂, CHCl₃, THF, 1,4-dioxane and DMF), (see the Supporting Information for details) however, only the use of DMF provided any conversion to the product, indicating that solvent polarity may be significant for the success of this transformation. A control reaction in DMF in the absence of the catalyst, however, also led to comparable conversion to the product, consistent with the operation of a competitive Brønsted base-promoted reaction (see the Supporting Information for details). Taking MeCN as the optimal solvent, the use of eight different organic and inorganic bases was investigated (see the Supporting Information for details). Of those tested, Et₃N provided an improved yield of 98%, whilst maintaining comparable diastereo- and enantioselectivity (Entry 3). The use of alternative aryl esters was next probed, with pentafluorophenyl ester 2 and bis(trifluoromethyl)phenyl ester 3 giving amide products 7 and 8 in high yield, but with lower enantioselectivity than when using PNP ester 1 (Entries 4 and 5). The use of 2,4,6-trichlorophenyl ester 4 resulted in only 31% yield (Entry 6), which is consistent with previous studies in this field [65,72,76,79], and most likely reflects the increased steric hindrance of the aryloxide attenuating its nucleophilicity. Finally, using PNP ester 1, the catalyst loading could be reduced to 5 mol% with only a small drop in stereoselectivity (Entry 7), while heating the reaction to 40 °C provided a slight improvement in yield (Entry 8). The reaction could also be performed in the absence of a base (Entry 9), however, slightly lower yield was obtained and therefore, during investigation of the substrate scope, Et₃N was routinely used as an auxiliary base.

2.2. Reaction Scope and Limitations

Due to the low diastereoselectivity observed using 2,6-di-tert-butyl para-QM 5, the alternative use of 2,6-disubstituted para-QMs were investigated (Scheme 2). 2,6-Dimethyl, dibromo and diphenyl para-QMs 9–11 bearing a phenyl substituent at the exocyclic olefin were applied under optimized conditions. In all cases significantly lower conversion was observed (≤43%), and the amide products 14–16 were difficult to isolate. Based on analysis of the crude reaction mixture by ¹H NMR spectroscopy, the 2,6-dimethyl and dibromo-substituted analogues 14 and 15 were obtained with marginally improved diastereoselectivity (~70:30 dr), indicating that alternative substituents in these positions could prove beneficial if the products were isolable. Next, variation of the exocyclic substituent was probed. Incorporation of a methyl group at this position provided no improvement in dr, but the amide product 17 was isolated in a 50% yield and with moderate enantioenrichment for both diastereoisomers. Finally, incorporation of a 2-naphthyl substituent at this position was well tolerated, with amide 18 obtained in an 83% yield, 70:30 dr and excellent enantiocontrol (94:6 er) for the major diastereoisomer.
Table 1. Reaction optimization.

<table>
<thead>
<tr>
<th>Entry</th>
<th>6 (mol%)</th>
<th>Aryl Ester</th>
<th>Base</th>
<th>Yield (%)</th>
<th>dr (7:8)</th>
<th>er (7)</th>
<th>er (8)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>20</td>
<td>1</td>
<td>i-Pr₂NEt</td>
<td>82</td>
<td>55:45</td>
<td>97:3</td>
<td>94:6</td>
</tr>
<tr>
<td>2</td>
<td>0</td>
<td>1</td>
<td>i-Pr₂NEt</td>
<td>0</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>3</td>
<td>20</td>
<td>1</td>
<td>Et₃N</td>
<td>98</td>
<td>60:40</td>
<td>97:3</td>
<td>94:6</td>
</tr>
<tr>
<td>4</td>
<td>20</td>
<td>2</td>
<td>Et₃N</td>
<td>99</td>
<td>65:35</td>
<td>81:19</td>
<td>94:6</td>
</tr>
<tr>
<td>5</td>
<td>20</td>
<td>3</td>
<td>Et₃N</td>
<td>88</td>
<td>60:40</td>
<td>94:6</td>
<td>91:9</td>
</tr>
<tr>
<td>6</td>
<td>20</td>
<td>4</td>
<td>Et₃N</td>
<td>31</td>
<td>55:45</td>
<td>84:16</td>
<td>71:29</td>
</tr>
<tr>
<td>7</td>
<td>5</td>
<td>1</td>
<td>Et₃N</td>
<td>95</td>
<td>55:45</td>
<td>93:7</td>
<td>88:12</td>
</tr>
<tr>
<td>8</td>
<td>5</td>
<td>1</td>
<td>Et₃N</td>
<td>99</td>
<td>60:40</td>
<td>92:8</td>
<td>85:15</td>
</tr>
<tr>
<td>9</td>
<td>5</td>
<td>1</td>
<td>none</td>
<td>90</td>
<td>60:40</td>
<td>93:7</td>
<td>89:11</td>
</tr>
</tbody>
</table>

All reactions were conducted using each diastereoisomer through sequential treatment with i-Pr₂NEt followed by in situ (S)-TM·HCl, (5)-TM HCl, para-nitrophenoxide and benzylamine (Scheme 3). Of those tested, Et₃N was routinely used as an auxiliary base. All reactions were carried out on a 0.25 mmol scale; isolated yields are a mixture of diastereoisomers. 

Although structural variation of the para-QM provided marginal improvements in dr, the use of 2,6-di-tert-butyl para-QM 5 was considered most convenient for further investigations due to its stability and ease of synthesis, and the higher yields of product obtained from catalysis. To investigate if the dr obtained in these reactions was a manifestation of a kinetic or thermodynamic preference, isolation of diastereomeric PNP esters 19 and 20 and resubjection to catalysis conditions was attempted. Although the diastereomeric amides 7 and 8 had proved difficult to separate, the corresponding PNP esters 19 and 20 were chromatographically separable and displayed high stability. Epimerization studies were conducted using each diastereoisomer through sequential treatment with i-Pr₂NEt, (S)-TM·HCl 6, para-nitrophenoxide and benzylamine (Scheme 3). These experiments were followed by in situ ¹H NMR spectroscopic analysis and revealed no epimerization in either case. This indicates the dr obtained in the catalytic reaction most likely reflects the inherent diastereoselectivity of the transformation. Following separation of the diastereoisomers, the absolute configuration of the major diastereoisomer could also be confirmed as (2S,1'R) by single crystal X-ray crystallographic analysis [83,84]. Based on literature precedent, the (S)-configuration at C(2) was expected to be generated under catalyst-control, and therefore the absolute configuration of the minor diastereoisomer was predicted to be (2S,1'S).
Scheme 2. Scope: Variation of para-quinone methide—0.25 mmol scale; only the structure of the major diastereoisomer is shown; isolated yield is a mixture of diastereoisomers; dr determined by $^1$H NMR spectroscopic analysis of the crude reaction product mixture; er determined by CSP-HPLC analysis. [a] Product not isolable: conversion based on $^1$H NMR analysis of the crude reaction product mixture; ers could not be determined.

Scheme 3. Control studies and confirmation of absolute configuration of products. The majority of hydrogen atoms are omitted for clarity within X-ray crystal structure representation of 19.
Having established the stability of the PNP ester products and demonstrated the potential to separate the diastereoisomers by column chromatography, the scope of the catalytic transformation was investigated through variation of the PNP ester substrate. In each case, the PNP ester product diastereoisomers were at least partially separable, enabling unambiguous characterization. To test the applicability of the procedure, $p$-toly-substituted PNP ester product 19 was prepared on a larger scale (1.25 mmol) (Scheme 4). A combined 85% yield of both diastereoisomers was obtained, with comparable stereoselectivity to that observed when the reaction was conducted on an analytical scale (Table 1, Entry 7). The generality of the procedure was further probed using five electronically- and sterically-differentiated PNP esters. Introduction of an electron-donating 4-methoxy substituent was well tolerated, with PNP ester 26 obtained in quantitative yield, 60:40 dr and with high enantioselectivity for both diastereoisomers. Under the optimized conditions, the introduction of an electron-withdrawing 4-trifluoromethyl group resulted in low enantioselectivity (27: 63:37 er), which was attributed to a competitive Brønsted base-promoted background reaction. Consistent with this hypothesis, repeating the reaction in the absence of Et$_3$N, and using the free base of the isothiourea catalyst 6, provided PNP ester product 27 in 94% yield and significantly improved enantioselectivity (88:12 er$_{\text{maj}}$; 79:21 er$_{\text{min}}$). A similar effect was observed when using a 2-naphthyl-substituted PNP ester, with optimal enantioselectivity obtained in the absence of an auxiliary base (28: 91:9 er$_{\text{maj}}$; 85:15 er$_{\text{min}}$). Introduction of a sterically-imposing 1-naphthyl or a heteroaromatic thienyl substituent was also tolerated, with 29 and 30 obtained in excellent yield and with high enantioselectivity.

**Scheme 4.** Scope: Variation of para-nitrophenyl ester—0.25 mmol scale; only the structure of the major diastereoisomer is shown; isolated yields are given for the combination of diastereoisomers; dr determined by $^1$H NMR spectroscopic analysis of the crude reaction product mixture; er determined by CSP-HPLC analysis. [a] $1.25$ mmol scale. [b] Conducted at r.t., in the absence of Et$_3$N, and using free base of 6.
2.3. Proposed Mechanism

The mechanism of this transformation is proposed to begin with N-acylation of the free base isothiourea catalyst 6 by PNP ester 31 to generate the corresponding acyl ammonium para-nitrophenoxide ion pair 32 (Scheme 5). Subsequent deprotonation leads to \((Z)\)-ammonium enolate 33. Based on previous mechanistic studies [78,85], and the catalytic activity observed in the absence of an auxiliary base (Table 1, Entry 9), deprotonation is likely to be affected by the para-nitrophenoxide counterion. 1,6-Conjugate addition of ammonium enolate 33 to para-QM electrophile 34, followed by protonation, gives acyl ammonium intermediate 35. Finally, regeneration of catalyst 6, and concurrent release of product 37, is proposed to be facilitated by para-nitrophenoxide [59–79,86,87]. Although not essential for reactivity, the addition of Et₃N as an auxiliary base may be beneficial as a proton shuttle, and to maintain the isothiourea catalyst in its non-protonated form 6 [77,86]. The enantioselectivity of the transformation indicates the C–C bond forming event takes place on the Si-face of the ammonium enolate. This selectivity can be rationalized through preferential formation of the \((Z)\)-ammonium enolate [76–79,85], which is conformationally-restricted by an intramolecular 1,5-O···S interaction [61,88–104] and results in the phenyl stereodirecting group of the catalyst blocking the enolate Re-face. The observed poor diastereoselectivity can be tentatively rationalized by a simple stereochemical model that assumes a favored, open pre-transition state assembly where steric interactions are minimized about the forming C–C bond. Minimal differentiation between the aryl- and quinone substituents of the para-QM quinone leads to the two transition state assemblies 38 and 39 that give the major and minor diastereoisomers, respectively.

Scheme 5. Proposed mechanism (only the pathway for the formation of the major diastereoisomer is shown) and transition state assemblies. ArOH corresponds to either PNPOH or the reaction product.
3. Materials and Methods

3.1. General Procedure for the Enantioselective 1,6-Addition

In a flame-dried vial, the requisite para-quinoine methide (1.0 equiv.), aryl ester (1.5 equiv.), (S)-TM HCl (5 mol%), Et$_3$N (1.0 equiv.) and anhydrous MeCN (0.6 M) was added and stirred at r.t. for 24 h. The reaction was then quenched with benzylamine (5.0 equiv.) and stirred at r.t. for a further 12 h before being concentrated in vacuo. The residue was diluted with EtOAc (20 mL) and washed successively with 10% citric acid (20 mL × 1), aqueous NaOH (20 mL × 3) and brine (20 mL × 1). The organic layer was extracted, dried over MgSO$_4$ and the filtrate was concentrated in vacuo. The crude material was purified by flash silica column chromatography to give the desired product.

3.2. Representative Synthesis and Characterization of Compounds 7 and 8 (Entry 8)

Following the general procedure above, 4-benzylidene-2,6-di-t-tert-butylcyclohexa-2,5-dien-1-one 5 (74 mg, 0.25 mmol), 4'-nitrophenyl 2-(p-toly)acetate 1 (102 mg, 0.375 mmol), (S)-TM HCl 6 (3 mg, 5 mol%) and Et$_3$N (35 μL, 0.25 mmol) were dissolved in anhydrous MeCN (0.42 mL). The reaction mixture was stirred at 40 °C for 3 h before being quenched with benzylamine (137 μL, 1.25 mmol) at r.t. to give a crude mixture containing the title compound in 60:40 dr. The mixture was purified by flash silica column chromatography (petroleum ether/EtOAc, 85:15) to afford diastereoisomers 7 and 8 (60:40 dr) (132 mg, 99%) as a pale yellow solid.

mp 126–128 °C; [α]$_D^{20}$ +10.0 (c 1.0, CHCl$_3$); IR $\nu_{max}$ (film)/cm$^{-1}$ 3638 (O–H) 3304 (N–H), 2955 (C–H), 1645 (C=O); HRMS (ESI$^+$) $\text{C}_27\text{H}_{43}\text{NO}_2$ [M + H$^+$], found 534.3359, requires 534.3367 (−1.4 ppm).

Data for major diastereoisomer (7): Chiral HPLC analysis, Chiralpak AD-H (10% i-PrOH/hexane, flow rate 1.5 mL min$^{-1}$, 211 nm, 40 °C), $t_{R}$ 8.5 min and 29.3 min, 92:8 er; $^1$H NMR (500 MHz, CDC$_3$) δ$_H$: 1.42 (18H, s, (C′′′′)C(CH$_3$)$_3$), C(3′′′′)C(CH$_3$)$_3$, 2.24 (3H, s, C(4′′′′)CH$_3$), 3.90–4.05 (2H, m, C(2′′′′)H), 5.14 (1H, s, O–H), 6.71–6.76 (2H, m, Ar), 6.96–7.15 (2H, m, Ar), 7.34 (1H, t, J 7.6, Ar), 7.43–7.51 (1H, m, Ar), $^{13}$C[I] NMR (126 MHz, CDC$_3$) δ$_C$: 21.0 (C(4′′′′)CH$_3$), 30.4 (C(3′′′′)C(CH)$_3$)$_3$, 34.4 (C(3′′′′)C(CH)$_3$)$_3$, 43.6 (C(CH)$_3$)$_3$, 54.1 (C(1′′′′), 59.5 (C(2′′′′)), 124.7 (C(2′′′′), C(6′′′′)), 125.8 (C(4′′′′)), 128.2 (Ar), 127.2 (Ar), 127.3 (Ar), 128.0 (Ar), 128.4 (Ar), 128.5 (Ar), 128.6 (Ar), 129.0 (Ar), 133.8 (C(1′′′′)), 135.2 (C(1′′′′)), 135.6 (C(3′′′′), C(5′′′′)), 136.4 (C(4′′′′)), 137.9 (i-Ph), 142.4 (C(1′′′′)), 152.4 (C(4′′′′)), 172.1 (C(1′′′′)).

Selected data for minor diastereoisomer (8): Chiral HPLC analysis, Chiralpak AD-H (10% i-PrOH/hexane, flow rate 1.5 mL min$^{-1}$, 211 nm, 40 °C), $t_{R}$ 3.8 min and 18.6 min, 85:15 er; $^1$H NMR (500 MHz, CDC$_3$) δ$_H$: 1.27 (18H, s, (C′′′′′)C(CH)$_3$)$_3$, C(5′′′′)C(CH)$_3$)$_3$, 2.27 (3H, s, C(4′′′′′)CH$_3$), 3.90–4.05 (2H, m, C(2′′′′′)H), 4.50 (1H, dd, J 15.0, 6.8, CH$_3$Ph), 4.69 (1H, d, J 11.7, C(1′′′′′)H), 4.89 (1H, s, O–H), 5.69 (1H, t, J 5.6, NH), 6.81–6.84 (2H, m, Ar), $^{13}$C[I] NMR (126 MHz, CDC$_3$) δ$_C$: 21.0 (C(4′′′′′)CH$_3$), 30.2 (C(3′′′′′)C(CH)$_3$)$_3$, C(5′′′′′)C(CH)$_3$)$_3$, 34.2 (C(3′′′′′)C(CH)$_3$)$_3$, 43.4 (C(CH)$_3$)$_3$, 54.7 (C(1′′′′′)), 59.5 (C(2′′′′′)), 125.3 (C(5′′′′′)C(CH)$_3$)$_3$, C(6′′′′′)), 126.3 (C(4′′′′′)), 127.5 (Ar), 128.2 (Ar), 128.4 (Ar), 128.8 (Ar), 132.1 (C(1′′′′′)), 134.9 (C(3′′′′′), C(5′′′′′)), 135.5 (C(1′′′′′)), 136.4 (C(4′′′′′)), 138.2 (i-Ph), 143.6 (C(1′′′′′)), 151.7 (C(4′′′′′)), 172.0 (C(1′′′′′)).

4. Conclusions

An isothiourea-catalyzed enantioselective 1,6-conjugate addition of para-nitrophenyl (PNP) esters to para-quinoine methides (QMs) has been developed. Variation of the arylactic ester and para-QM substrates has provided a range of functionalized products in generally excellent yields and high enantiocontrol (up to 94:6 er). An inherent limitation of the method is that the products were routinely obtained in ~60:40 dr. This diastereoselectivity was shown to arise from kinetic control, but was relatively insensitive to changes in reaction conditions and structural variation of the substrates. Although the dr could
not be improved, the diastereoisomeric PNP ester products could be separated by column chromatography. The success of this catalytic methodology is proposed to rely upon the para-nitrophenoxide, expelled during N-acylation of the catalyst, to facilitate catalyst turnover and release the product. Current work in our laboratory is focused on further applications of using in situ-generated aryloxides to promote catalyst turnover in Lewis base catalysis.

Supplementary Materials: Full experimental procedures, characterization data, NMR spectra and HPLC chromatograms for all new compounds, as well as crystallographic data for product 19 (CCDC 1992504) are available online.

Author Contributions: Conceptualization, A.D.S. and J.N.A.; investigation, J.N.A.; W.C.H.; C.M.; X-ray crystallographic analysis, A.M.Z.S.; writing—original draft preparation, M.D.G.; writing—review and editing, A.D.S and all authors; supervision, D.S., S.N. and A.D.S.; funding acquisition, A.D.S. All authors have read and agreed to the published version of the manuscript.

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Sample Availability: Samples of the compounds are available from the authors on request.

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84. The relative and absolute configuration of all other products was assigned by analogy, with consistent diagnostic 1H NMR signals of the proton at C(2) and C(1'R) stereocentres, indicating the same major diastereoisomer in each case. [CrossRef]


87. Catalyst turnover could also be envisaged through intramolecular nucleophilic displacement by the pendant phenol to give a cyclobutanone intermediate via a formal [2+2]-cycloaddition, which may then be ring-opened by para-nitrophenoxide to give the final product. Based on the high steric congestion of this proposed cyclobutanone intermediate, the mechanism given in Scheme 5 is considered more probable. [CrossRef]


