Star Polymers Reduce Islet Amyloid Polypeptide Toxicity via Accelerated Amyloid Aggregation

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Supporting Information

ABSTRACT: Protein aggregation into amyloid fibrils is a ubiquitous phenomenon across the spectrum of neurodegenerative disorders and type 2 diabetes. A common strategy against amyloidogenesis is to minimize the populations of toxic oligomers and protofibrils by inhibiting protein aggregation with small molecules or nanoparticles. However, melanin synthesis in nature is realized by accelerated protein fibrillation to circumvent accumulation of toxic intermediates. Accordingly, we designed and demonstrated the use of star-shaped poly(2-hydroxyethyl acrylate) (PHEA) nanostructures for promoting aggregation while ameliorating the toxicity of human islet amyloid polypeptide (IAPP), the peptide involved in glycemic control and the pathology of type 2 diabetes. The binding of PHEA elevated the $\beta$-sheet content in IAPP aggregates while rendering a new morphology of “stelliform” amyloids originating from the polymers. Atomic molecular dynamics simulations revealed that the PHEA arms served as rodlike scaffolds for IAPP binding and subsequently accelerated IAPP aggregation by increased local peptide concentration. The tertiary structure of the star nanoparticles was found to be essential for driving the specific interactions required to impel the accelerated IAPP aggregation. This study sheds new light on the structure–toxicity relationship of IAPP and points to the potential of exploiting star polymers as a new class of therapeutic agents against amyloidogenesis.

INTRODUCTION

Type 2 diabetes mellitus (T2D) is a metabolic disease affecting 5% of the global population.† Extensive research indicates that a major factor in the development and pathogenesis of T2D is dysfunction of human islet amyloid polypeptide (IAPP), a 37-residue peptide cosecreted with insulin from pancreatic $\beta$-cells, which undergoes fibrillation to form amyloid plaques found in 90% of T2D patients.‡,§ The toxic IAPP aggregation products are also capable of eliciting systemic damage in T2D patients with evidence of cardiac dysfunction and neurological deficits mediated by IAPP deposition increasing the burden of disease.$^*$,¶ Concordantly, there is a crucial need for the development of treatment agents that are capable of mitigating IAPP-associated toxicity in vivo to reduce the morbidity of T2D and prevent its development in prediabetics.

Aggregation inhibition with the use of small molecules as well as metal, carbon, and polymeric nanoparticles (NPs)$^6$–$^9$ has been a major strategy against amyloid-mediated toxicity. Polymeric NPs, specifically, have been explored as protein aggregation inhibitors utilizing their tunable hydrophobicity as well as their capacity for initiating H-bonding.$^{10–13}$ For example, antiprion activity has been demonstrated by phosphorus dendrimers, maltose-based glycodendrimers (mPPI), poly(propyleneimine) PPI, and poly(ethylenimine) hyperbranched polymers.$^{12,20}$ Of the myriad forms of polymeric NPs, hyperbranched polymers and dendrimers have demonstrated strong efficacies as antiamyloid agents,$^{11,12,15,21–26}$ though anti-IAPP applications have only been explored recently. PPI glycodendrimers and lysine dendrimers have been investigated as anti-$A\beta$ aggregation agents,$^{22,23}$ and hyperbranched PEG-based polymers with a dopamine moiety were found to be capable of inhibiting $\alpha$-
synuclein (a5) aggregation. Anionic low-generation dendrimers have been shown to modulate IAPP fibrillation and associated toxicity, and inhibition of IAPP fibrillation and toxicity in vitro and ex vivo has recently been demonstrated by our team with OH-terminated polyamidoamine dendrimers (PAMAM-OH).

Curiously, the amyloidogenesis of melanocyte protein Pmel17 in the human system is entirely nonpathogenic. The rapid fibrillation of Pmel17, which transitions from monomeric form to mature amyloid fibrils within 3 s, is a cytoprotective mechanism, namely, through reducing the half-life of toxic intermediate products (i.e., oligomers and protofibrils) to favor the formation of nontoxic, mature amyloid fibrils. Indeed, this effect has been shown to extend to pathogenic amyloids; mice overexpressing the Alzheimer’s-related amyloidogenic peptide amyloid-b (Aβ) with the “Artic” mutation, correlated with the acceleration of Aβ fibrillation, demonstrated higher plaque loading with lower or negligible impact on behavioral function comparable to mice expressing wild-type Aβ. The concept of fibrillation promotion, rather than inhibition, thus provides a biomimetic and perhaps counterintuitive strategy in the mitigation of amyloid cytotoxicity.

Amyloid aggregation promotion as a strategy for mitigating cytotoxicity has thus far only been reported for a selected few small molecules. Specifically, high-throughput screening identified aromatic small molecules capable of promoting Aβ fibrillation to provide a cytoprotective effect, including the orcein-related molecule O430 and the compound 2002-H20.29 The rapidity of fibrillation promotion, rather than inhibition, thus provides a biomimetic and perhaps counterintuitive strategy in the mitigation of amyloid cytotoxicity.

EXPERIMENTAL METHODS

Materials. 2-Hydroxyethyl acrylate (HEA) was purchased from Sigma-Aldrich and dehydrated by passing through a column of alumina. 2,2′-Dibenzyl trithiocarbonate (DBTC), N,N′-methylenebis(acrylamide) (X) was purchased from Sigma-Aldrich. Azobis(isobutyronitrile) (AIBN) was purified by recrystallization from methanol before use. Dimethyl sulfoxide (DMSO) was purchased from Merck Millipore and used as received. Human islet amyloid polypeptide monomers (IAPP; disulfide bridge: 2–7; MW: 3,906; 37 residue: KCNTATCATQQLNFLVHSNFPAGILSTNVSNTY; >95% pure by HPLC) were obtained in lyophilized powder form from AnaSpec and were made up to a 200 μM stock immediately prior to an experiment or allowed to fibrillate at 25 °C for >5 days to produce mature IAPP amyloids. All materials were weighed out on a Cubis MSE balance (Sartorius, 0.01 mg resolution) and made up fresh in MilliQ water prior to experiments unless otherwise specified. Thiodialin T (ThT) dye (Sigma-Aldrich) was prepared fresh for each experiment at a 250 μM stock solution. Propidium iodide (PI) dye stock solution (1 mg/mL in water) was stored at −20 °C.

RAFT Synthesis of Poly(2-hydroxy ethyl acrylate) Stars. Synthesis of Poly(2-hydroxy ethyl acrylate) (p(HEA)) Homopolymer. Three homopolymers of PHEA were prepared by RAFT polymerization with molecular weights of 4,000, 8,000, and 16,000 g/mol. The synthesis of PHEA 1 (Mn = 4,000 g/mol) was carried out using the following stoichiometry: [DBTC]:[HEA]:[AIBN] = 1:38:0.1. In brief, 2-hydroxyethyl acrylate (4.0 g, 0.034 mol), DBTC RAFT agent (0.27 g, 9.12 × 10^-4 mol), AIBN (15.4 × 10^-3 g, 9.38 × 10^-3 mol), and DMSO (26 mL) were placed into a 50 mL round-bottom flask equipped with a magnetic stirrer bar and capped with a rubber septum. The reaction mixture was deoxygenized for 1 h at 0 °C with nitrogen gas. The sealed deoxygenated reaction vessel was placed in a preheated oil bath at 70 °C, and the polymerization was carried for 4 h with stirring. Polymerization was stopped by placing the vessel in ice to cool. The product was analyzed by 1H NMR and GPC. The monomer conversion was determined to be approximately 88% by NMR, thus resulting in a Mn of 4,141 g/mol. By integrating the peaks associated with the benzyl group (7.1–7.4 ppm) and the hydroxyl group in the PHEA repeat unit (δ = 4.8 ppm), the Mn can be calculated to be 4,374 g/mol. Per arm, there are 32 units of the PHEA monomer because the homopolymer constitutes 64 units. The molecular number and polydispersity index were determined by GPC to be 8,276 Mn and 1.30, respectively.

A similar procedure was employed for the synthesis of PHEA 2 (Mn = 8,000 g/mol) using the following stoichiometry: [DBTC]:[HEA]:[AIBN] = 1:75:0.1. In brief, 2-hydroxyethyl acrylate (4.0 g, 0.034 mol), DBTC RAFT agent (0.13 g, 4.58 × 10^-4 mol), AIBN (7.2 × 10^-3 g, 4.38 × 10^-3 mol), and DMSO (26 mL) were placed into a 50 mL round-bottom flask equipped with a magnetic stirrer bar and capped with a rubber septum. The reaction mixture was deoxygenized for 1 h at 0 °C with nitrogen gas. The sealed deoxygenated reaction vessel was placed in a preheated oil bath at 70 °C, and the polymerization was carried for 4 h with stirring. Polymerization was stopped by placing the vessel in ice to cool. The product was analyzed by 1H NMR and GPC. The monomer conversion was determined to be approximately 85% by NMR, thus resulting in a Mn of 4,141 g/mol. By integrating the peaks associated with the benzyl group (7.1–7.4 ppm) and the hydroxyl group in the PHEA repeat unit (δ = 4.8 ppm), the Mn can be calculated to be 4,374 g/mol. Per arm, there are 32 units of the PHEA monomer because the homopolymer constitutes 64 units. The molecular number and polydispersity index were determined by GPC to be 14,688 Mn and 1.24, respectively.

The synthesis of PHEA 3 (Mn = 16,000 g/mol) was achieved in a similar method as above using the following stoichiometry: [DBTC]:[HEA]:[AIBN] = 1:144:0.1. In brief, 2-hydroxyethyl acrylate (4.91 g, 0.042 mol), DBTC RAFT agent (0.085 g, 2.93 × 10^-4 mol), AIBN (5 × 10^-3 g, 3.04 × 10^-3 mol), and DMSO (32 mL) were placed into a 50 mL round-bottom flask equipped with a magnetic stirrer bar and capped with a rubber septum. The reaction mixture was deoxygenized for 1 h at 0 °C with nitrogen gas. The sealed deoxygenated reaction vessel was placed in a preheated oil bath at 70 °C and the polymerization was carried for 6 h with stirring. Polymerization was terminated by placing the vessel in ice to cool. The product was...
analyzed by $^1$H NMR and GPC. The monomer conversion was determined to be approximately 91% by NMR, thus resulting in a $M_n$ of 15,553 g/mol. By integrating the peaks associated with the benzyl group (7.1–7.4 ppm) and the hydroxyl group in the PHEA repeat unit (δ = 4.8 ppm), the $M_n$ (NMR) is found to be 15,961 g/mol. The molecular number and polydispersity index were determined by GPC to be 26,676 $M_n$ and 1.24, respectively.

**Synthesis of PHEA Star.** A typical PHEA star synthesis is carried out as follows. The synthesis of PHEA star was conducted using the following stoichiometry: [PHEA 2: [HEA]: [ABN]: [X]L] = 1:12:0.4:1. In brief, a 1.86 mL aliquot (250 mg equivalent) of the solution above, ABN (1.60 mg, 9.76 × 10$^{-3}$ mol), and 77.09 mg 0.5 mol of DMAc were added into a glass vial equipped with a magnetic stirrer bar and capped with a rubber septum. The reaction mixture was deoxygenated for 30 min at 0 ºC with nitrogen gas. The sealed deoxygenated reaction vessel was placed in a preheated oil bath at 70 ºC, and the polymerization was carried for 24 h with stirring. Polymerization was terminated by placing the vessel in ice to cool.

Purification was conducted by dialysis against water with a MWCO of 14 kDa for 2 days and then lyophilization. The product was analyzed by GPC, and its molecular number and polydispersity index were determined to be 85,500 $M_n$ and 2.18, respectively. With the given $M_n$ values from GPC, the star was calculated to have 12 arms. In theory, the molecular weight of the star is 47,295 g/mol.

**Analysis Methods.** $^1$H Nuclear Magnetic Resonance Spectroscopy. $^1$H NMR spectra were recorded at 400 MHz on a Bruker UltraShield 400 MHz spectrometer running Bruker Topspin, version 1.3. Spectra were recorded in DMSO-d$_6$.

**Gel Permeation Chromatography (GPC).** GPC was performed using a Shimadzu modular system comprised of a SIL-20AD automatic injector, a RID-10A differential refractive-index detector, and a 50 × 0.8 mm guard column followed by three KF-805L columns (300 × 7.8 mm, bead size: 10 µm, guard column followed by three KF-805L columns (300 × 7.8 mm, bead size: 10 µm). Molar mass determination was carried out as follows. The synthesis of PHEA star was conducted using the following stoichiometry: [PHEA 2: [HEA]: [ABN]: [X]L] = 1:12:0.4:1. In brief, a 1.86 mL aliquot (250 mg equivalent) of the solution above, ABN (1.60 mg, 9.76 × 10$^{-3}$ mol), and 77.09 mg 0.5 mol of DMAc were added into a glass vial equipped with a magnetic stirrer bar and capped with a rubber septum. The reaction mixture was deoxygenated for 30 min at 0 ºC with nitrogen gas. The sealed deoxygenated reaction vessel was placed in a preheated oil bath at 70 ºC, and the polymerization was carried for 24 h with stirring. Polymerization was terminated by placing the vessel in ice to cool.

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Dynamic Light Scattering (DLS). DLS was carried out on a Malvern Zetasizer Nano ZS Series running DTS software (laser, 4 mW, λ = 633 nm, angle 173º). The polydispersity index (PDI) used to describe the average diameters and size distribution of prepared stars was determined via a cumulants analysis of the measured intensity autocorrelation function using the DTS software. Samples were filtered through 0.45 µm PTFE filters before injection. The GPC calibration was performed with narrow-polydispersity polystyrene standards ranging from 500 to 2 × 10$^6$ g mol$^{-1}$.

**Fourier Transform Infrared Spectroscopy-Attenuated Total Reflectance (FTIR-ATR).** ATR-FTIR measurements were performed using a Shimadzu IRT-100. Fourier transform infrared spectrometer with a GladiATR 10 single reflection ATR accessory. Spectra were obtained in the midinfrared region of 4000–600 cm$^{-1}$ at a resolution of 8 cm$^{-1}$ (512 scans) and analyzed using LabSolution IR software.

**Thioflavin T Assay.** Aliquots of IAPP (final concentration of 25 µM), ThT dye (25 µM), and PHEA polymers (5, 10, or 0.2 µM) were added directly to wells of a black/clear bottom 96-well plate (Corning) and mixed thoroughly. The final well volume of 100 µL was made up using Milli-Q water where necessary. The plate was run on a Flexstation 3 plate reader (Molecular Devices) with samples excited at 440 nm and the emission read at 485 nm every 5 min for a total of 14 h (169 readings).

**Circular Dichroism (CD) Spectroscopy.** Experiments were performed on a Chirascan CD spectrometer (Applied Photophysics) with spectra read from 190 to 260 nm. Prior to sample loading, a baseline with no cuvette was run. Then, 300 µL of 25 µM IAPP in Milli-Q water alone or in the presence of PHEA (5 µM), was placed in a cuvette with a 0.1 cm path length, and CD analysis was run at 0, 2.5, and 24 h time points. Between samples, cuvettes were washed more than 5× with distilled water. Reads are an average of 3 repeats. Raw data were offset to zero and normalized against the spectra of Milli-Q water for IAPP spectra and against PHEA alone for IAPP-PHEA mixed samples. Data were then deconvoluted with CDNN software to give a final relative percentage content of secondary structure.

Transmission Electron Microscopy. Samples were placed in Eppendorf tubes at a final IAPP concentration of 25 µM and incubated for 24 h at 25 ºC. An aliquot (10 µL) was placed on 400 mesh carbon-coated Formvar copper grids (ProSciTech) that were glow-discharged to promote hydrophilicity. Samples were adsorbed onto the grid for 60 s, then drawn off using filter paper. Grids were washed twice with 10 µL of Milli-Q water. Five microliters of 1% uranyl acetate (in water) was then utilized to utilize twice-staining by touching one droplet and immediately drawing the stain off, and then placing the grid atop the second droplet to stain for 15 s. TEM images were obtained on a Tecnai TF20 transmission electron microscope (FEI) with an UltraScan 1000 (2k × 2k) CCD camera (Gatan).

**Determination of Fibril Morphology.** Fibril tracking and analysis were performed with software FiberApp$^{16}$ to determine the morphology and mesoscopic parameters of persistence length ($\lambda$) and contour length ($L$) of IAPP fibrils. FiberApp was developed from statistical physics and enables structural analysis of tubular and macromolecular objects. The persistence length $\lambda$ reflects the stiffness of a polymer and is mathematically defined via the bond correlation function (BCF) in 3D or 2D as the length over which angular correlations in the tangential direction decrease by a factor of e$^{-1}$. Here, the $\lambda$ values of IAPP fibrils were estimated using the BCF, mean-squared end-to-end distance (MSED), and mean-squared midpoint displacement (MSMD) methods and presented as averaged values determined by the three methods. The contour length corresponds to the end-to-end length of a polymer along its physical contour. The values of persistence length and contour length were obtained based on statistical analysis of 1,243 fibrils.

**Cell Culture and Viability.** Insulin-producing βTC6 cells (ATCC) were cultured in complete DMEM (ATCC; 15% FBS). For viability assays, a 96-well plate (Corning) was coated with 70 µL of 70 µg/mL poly-γ-lysine for >10 min; then, the wells were washed 3× in 100 µL of HBSS. Cells were seeded at a density of ~50,000 cells per well in 200 µL of complete media and incubated at 37 °C in 5% CO$_2$ for 3 days. Fresh IAPP or mature IAPP amyloids (200 µM stock) were preincubated together with PHEA polymers (40, 8, and 3.2 µM stocks) at a 1:1 v/v ratio for 24 h at room temperature. Prior to cell treatment, the media was aspirated, and the wells were washed 1× in 100 µL of HBSS. One micromolar propidium iodide (PI; AnaSpec) dye solution was made up in complete media containing 1% penicillin/streptomycin, and 150 µL aliquots were added to each of the wells. The cells were returned to the incubator for 30 min to equilibrate with the dye solution. Each sample treatment was then added to the wells in triplicate, in addition to an IAPP control made up fresh immediately prior to adding to the wells (final well volume: 200 µL; final IAPP concentration: 25 µM). The cells were imaged on an Operetta High-Content Imaging System (PerkinElmer) utilizing standardized excitation/emission settings for PI with images of five areas within a single well taken every hour for 24 h. Total cell counts per well were estimated using phase-contrast mapping within sampling areas. Cell death over time was expressed as %PI positive cells within the total cell count.

**Ex Vivo Viability.** C57BL/6 male mice (age 10–14-week-old) were maintained at St. Vincent’s Institute animal care facility on a 12 h light-dark cycle in a temperature-controlled room and obtained food and water ad libitum. Uniformly sized mouse islets from C57BL/6 mice were handpicked into 1 cm Petri dishes containing 1 mL of 25 µM IAPP, 5 µM PHEA, or a combination of both and cultured for 48 h. At the end of the culture period, islets were dispersed with trypsin and resuspended in 250 µL of hypotonic buffer containing 50 µg/mL of propidium iodide, which stained nuclear DNA. The cells were analyzed by fluorescence-activated cell sorter (FACS), and cell death was identified by their subdiploid DNA content as previously described. The study was conducted at St Vincent’s Institute (Melbourne, Australia) following the guidelines of the Institutional Animal Ethics Committee.
Statistics. Where applicable, data were analyzed using a one-way ANOVA with Tukey’s correction with \( p < 0.05 \) considered statistically significant.

**DMD Simulations and Simulation Setup.** DMD is a special form of molecular dynamics, where discrete step functions instead of continuous functions are used to mimic the constraints. The united-atom representation with all polar hydrogen and heavy atoms was used to model IAPPs and PHEA polymers. An adapted Medusa force field\(^{35}\) with an implicit solvent model was used to describe the nonbonded interatomic interactions, including van der Waals, solvation, hydrogen bonding, and electrostatic terms. The Debye–Hückel approximation with a Debye length of \( \sim 10 \) Å was applied to capture the screened electrostatic interactions. Anderson’s thermostat was used to maintain temperature, which was fixed at 300 K in all simulations. The DMD program is freely available to academic users at the Molecules in Action Web site (http://moleculesinaction.com), and all simulation parameters can be obtained upon request. Given the stochastic nature in the dynamics of a multimolecular system and the nucleation-dependent aggregation kinetics, we performed multiple independent long simulations with different starting configurations (e.g., randomized velocities, intermolecular distances and orientations) to ensure sufficient sampling and avoid potential bias associated with initial configurations. Thermodynamic and kinetic prosperities were then obtained by analyzing all the independent runs for each molecular system.

All the PHEA model structures were constructed with the Avogadro\(^{39}\) molecular builder software and energy minimized with the MMFF94s force field.\(^{40}\) MedusaScore,\(^{41}\) an extension of the Medusa force field,\(^{42}\) was adapted to model the polymers in addition to IAPP. The MedusaScore was parametrized on a large set of ligands and was transferrable to different molecular systems. The predictive power of MedusaScore has been validated in various benchmark studies, including recent community structure–activity resource (CSAR) blind ligand–receptor docking prediction exercises.\(^{43,44}\)

For each 2-arm PHEA (Figure S3A), we performed 20 independent simulations at 300 K with different starting configurations. Each independent simulation lasted 300 ns, and thus, an accumulative 6 \( \mu \)s simulation was obtained for the polymer model. We used the last half of all simulations and computed the radius of gyration \( (R_g) \) values of the modeled polymers. For the 8-arm PHEA (Figure S3C), 20

![Figure 1](image_url)
independent simulations were performed at 300 K with different starting configurations, each of which lasted 200 ns. The IAPP (PDB ID: 2L86) structure was obtained from PDB. Counterions Cl\(^-\) were introduced to achieve a neutral charge condition if necessary. For systems containing six IAPPs (with and without a 6-arm PHEA as shown in Figure S3B), the peptide concentration was maintained by fixing the dimension of the simulation box as 120 Å, and periodic boundary conditions were applied. For each of the multimolecular systems, 20 independent simulations starting with different intermolecular distances and orientations were performed at 300 K, and each run lasted 100 ns. Secondary structure analyses were performed using the dictionary secondary structure of protein (DSSP) method. For each snapshot structure, the secondary structure, such as helix, sheet, coil, and turn, for each residue was obtained. An empirical sigmoidal function

\[
y = \frac{(I_{\text{max}} - I_{\text{min}})}{(1 + \exp(-k(t - t_0)))} + I_{\text{min}}
\]

was adopted to fit the kinetics of the total number of residues in the β-sheet conformation, where fitting parameters \(A, B, t_0\) and \(k\) corresponded to the max and min values of aggregation, the midpoint time of aggregation, and the elongation rate, respectively. The lag time was determined as

\[
t_{\text{lag}} = t_0 - \frac{2}{k}
\]

In the potential of mean force (PMF) calculation, normalized sigmoidal function, \(Q_{\text{fibrillation}} = 1/(1 + \exp(-k(t - t_0)))\), was used to quantify the extent of fibrillation for each independent simulation. For a given snapshot, the distribution of IAPP oligomers was analyzed, where any two peptides interconnected by at least one intermolecular heavy atom contact (the cutoff of 0.55 nm) was defined to belong to an oligomer. The size of an oligomer, \(n_{\text{oligomer}}\), was defined by the number of IAPP peptides forming the aggregate. The two-dimensional PMF (or effective free energy) was computed according to

\[
\text{PMF} = -K_B T \ln P(n_{\text{oligomer}}, Q_{\text{fibrillation}})
\]

where \(K_B\) is the Boltzmann constant, \(T\) corresponds to the simulation temperature 300 K, and \(P(n_{\text{oligomer}}, Q_{\text{fibrillation}})\) is the probability of finding a peptide in an oligomer with the size of \(n_{\text{oligomer}}\) and the extent of fibrillation \(Q_{\text{fibrillation}}\) at the time.

\section*{RESULTS AND DISCUSSION}

\subsection*{Synthesis and Characterization of PHEA Stars.} PHEA stars were synthesized with constitutive elements similar to those observed in small molecule aggregation promoters. Specifically, the stars were designed to incorporate aromatic rings (on the distal end of the star arms) and pendant hydroxyls on the side chains of the arms. To actualize these design criteria, we chose SS'-dibenzyldithiobenzoate (DBTC) as the RAFT agent and employed in the polymerization of 2-hydroxethyl acrylate (Figure 1A). The polymerization was carried out in DMSO at 70 °C using AIBN as the radical initiator. To provide a suitable array of arm molecular weights for the subsequent star formation, we first synthesized three PHEA homopolymers with varying degrees of polymerization to approximately 80–90% conversion. The \(M_n\) of the homopolymers was determined via \(^1\text{H}\) NMR analysis by comparing integrals for the peaks at \(\delta = 7.1–7.4\) ppm (corresponding to the benzyl leaving group of the RAFT agent, i.e., the polymer end group) with the integral for the peak corresponding to the hydroxyl group in the PHEA repeat unit (\(\delta = 4.8\) ppm). These results agreed well with the theoretical molecular weights determined from the ratio of the monomer to RAFT agent (see Table S1). GPC analysis of the various PHEA homopolymers revealed unimodal peaks with an acceptable dispersity of 1.2–1.3, although of course the \(M_n\) values from GPC differed considerably from those determined above due to calibration of the GPC against polystyrene standards.

The solutions of arm polymers (including unreacted HEA) were then used to form the PHEA stars by adding methylene(bis(acrylamide)) at various molar ratios and a further aliquot of AIBN and then heating at 70 °C for 24 h. Samples were taken periodically during the reaction with a further aliquot of AIBN injected after 12 h. A library of different core-cross-linked stars were formed with the results tabulated in Table S2. It is worth noting that the symmetrical nature of the RAFT agent used (DBTC) gave rise to arm polymer in which the thiocarboxylthio moiety was in the middle of the polymer chain. As such, subsequent introduction of the difunctional monomer (to facilitate formation of the cross-linked core) occurred in the middle of the arm polymer. The consequence of this architecture is that the final stars had an arm molecular weight that was half the value of the starting “arm” material. To our knowledge, this is the first time that this approach (i.e., the use of a symmetrical RAFT agent such as DBTC) has been employed in the preparation of star polymers. The resulting star polymers were analyzed by gel permeation chromatography to identify the best conditions for star formation. For all systems, there was a shift in the GPC trace for the PHEA homopolymer to shorter retention times, reflecting successful chain extension, with this shift typically most pronounced for higher ratios of cross-linker to polymer (Figure 1B; for \([P]:[M]:[X] = 1:12:16\) and \(M_n(GPC)_{\text{arm}} = 14,500\) g mol\(^{-1}\)). Estimation of approximate arm number could be made by dividing the \(M_n(GPC)\) for the star by 1/2 \(M_n(GPC)\) for the arm. The factor of 1/2 is introduced into this equation because of the symmetry of the chain transfer agent, as noted above. For each series, higher ratios of cross-linker led to higher arm numbers. It should be noted, however, that these values are only indicative of the true arm number due to the potential underestimation of molecular weight when applying GPC to hyperbranched materials. Moreover, the indicative nature of this approach is clearly evident in that the arm number should only ever be even given the symmetrical nature of the RAFT agent. Importantly, the use of DBTC as a RAFT agent for the HEA polymerization and the subsequent formation of star from DBTC-derived PHEA led to somewhat broader molecular weight distributions than have been achieved using other star polymer systems.\(^{33}\) Although there is evidence in the literature that varying the solvent for star formation can improve arm incorporation and minimize polydispersity,\(^{33}\) the scope for optimization of the current system was limited by the intransigent solubility of the stars in most solvents: we observed that the resulting materials were soluble only in DMSO or water. Even then, some difficulty was encountered when attempting to redisperse the star in these solvents after lyophilization. We attribute this effect to hydrogen bonding between the arms and as such allowed a minimum of 72 h for the polymer to equilibrate after redispersing in water.

From the GPC data, the star with the clearest shift to higher molecular weight accompanied by maintenance of a relatively unimodal molecular weight distribution was observed for the system where \(M_n(GPC)_{\text{arm}} = 14,500\) g mol\(^{-1}\) and \([P]:[M]:[X] = 1:12:16\). Analysis of this material by dynamic light scattering (Figure 1C) indicated a number-average hydrodynamic radius of approximately 9.8 nm for the stars when dispersed in water, which is consistent with other water-soluble star polymers synthesized via RAFT polymerization. Spectroscopic evaluation of these same PHEA stars was also conducted with the recorded \(^1\text{H}\) NMR spectrum confirming that the desired
aromatic and hydroxyl moieties were present in the final polymer structure (Figure 2A). FTIR analysis before and after star formation provided further evidence of the high proportion of OH groups in the star in addition to demonstrating the presence of the methylene(bis(acrylamide)) linking groups in the star core (Figure 2B). Specifically, the emergence of peaks associated with amide C=O stretch (1,645 cm⁻¹) and amide N−H stretch (1,547 cm⁻¹) provide clear evidence that the amide cross-linking groups were successfully incorporated into the star structure.

Modulation of IAPP Fibrillization by PHEA Polymers. The thioflavin T (ThT) assay utilizes the amyloidophilic ThT dye to provide a measurement of both the extent and kinetics of amyloid fibrillation over time. The control IAPP was shown to fibrillate with a nucleation period of ~2 h followed by an exponential period of ~12 h before reaching saturation at 14 h (Figure 3A). Incubation of PHEA with IAPP at molar concentrations of 1:5 and 1:25 had a promotional effect on IAPP fibrillation both for the star NPs and their constituent subunits, referred to as “arms” (Figure S1), with the largest increase in ThT fluorescence observed at 1:5. Following the trend of increasing fibrillation with increasing concentration of PHEA, the energetically unfavorable nucleation period was also shown to significantly decrease with increasing concentration of PHEA, falling from ~3.5 h in the IAPP control to less than 30 min with the highest concentration of PHEA. The reduction in IAPP lag time was also demonstrated through circular dichroism (CD) spectroscopy, wherein the presence of PHEA to IAPP at the 1:5 ratio notably promoted the amyloidogenic conversion of random coil content to β-sheets (Figures 3B and Figure S3). Over 2.5 h, β-sheet conversion in IAPP:PHEA 1:5 (25−34%) was 2.3× more rapid than that of IAPP alone (29−33%), and concordantly, IAPP contained 25% higher β-sheet content at 24 h in the presence of PHEA (55%) compared to that of IAPP alone (44%). This shift in prototypical IAPP aggregation kinetics, in addition to the
promotion of fibrillization overall, presents a case for rapid local sequestration of IAPP seeds by PHEA, whereas a lower PHEA concentration ratio would isolate aggregation "hotspots" and thus limit fibril elongation but promote coaggregation and thus enhance fibrillization at higher PHEA concentrations.

Stelliform Amyloid Formation and Amyloid Remodeling by PHEA. Transmission electron microscopy (TEM) imaging complemented ThT and CD analyses for fibrillating IAPP, allowing further analysis of persistence and contour length of amyloid fibrils generated after 24 h in aqueous solution in the presence and absence of PHEA. After 24 h, IAPP amyloidogenesis reached the saturation phase, and long, semiflexible fibrils can be observed by TEM with some shorter species still present (Figure 4A). Once fibril elongation and 3D cross-linking occurred at >5 days of amyloidogenesis, amyloids formed in-solution hydrogels, and generally, shorter species were absent. Stelliform amyloids were observed when PHEA was incubated with IAPP at a 1:5 molar ratio (Figure 4A). These amyloids were characterized by a central nucleation “core”, ranging from smaller clusters of 50–150 nm to micrometers in diameter. Fibrils of low persistence (average of 891.9 nm compared with that of 2,885 ± 60 nm for the IAPP control) and contour length (<1,350 nm) were additionally observed radiating out from the core, forming the full stelliform structure of ∼0.5 μm in diameter for smaller cores and micrometers in diameter for larger cores with some macroscopic aggregates visible in solution (Figure 4A).

With lower concentrations of PHEA, the fibrillization products generally trended toward matching the structural morphology of IAPP alone (Figure 4B). IAPP amyloid fibrils with significantly reduced contour lengths were produced with increasing PHEA concentration, indicating polyphenol-like stabilization of growing fibrils through H-bonding; hydrophobic and π−π interactions by PHEA may have terminated fibril elongation, resulting in a fibril population with predominantly low contour lengths. As fibril elongation is mediated by amyloid seeds, the extensive exponential periods observed in the ThT assay (Figure 3A) could be indicative of the PHEA-IAPP interaction.
complex rapidly sequestering amyloid seeds to render large populations of shorter fibrils, thus depleting the available seeding population to perform elongation. Interestingly, however, the shift in fibril persistence length induced by stars versus arms varied greatly; stars induced softer fibrils, whereas those generated in the presence of arms were notably stiffer. The capacity of PHEA to remodel IAPP amyloids was also assessed after coincubation for 24 h and demonstrated a similar trend. Amyloids were reduced in persistence length upon exposure to PHEA stars with the most notable effect seen as soft, limp fibrils “bundling” at PHEA:IAPP 1:5. Conversely, negligible amyloid remodeling was observed in the presence of PHEA arms. These observations could be attributed to the differences in morphology and surface physicochemical properties of the stars and arms.

Stelliform IAPP Amyloids Are Cytoprotective in Vitro and ex Vivo. IAPP-mediated cytotoxicity was assessed in an insulin-producing pancreatic β-cell line over a 24 h period (Figure 5A) and ex vivo in mouse islets after 48 h (Figure 5B). PHEA stars were completely biocompatible at all concentrations tested. In vitro, IAPP alone typically began to induce cytotoxicity at ~6 h post-treatment with cell death progressing exponentially up to the 20–24 h mark to an end point toxicity value of 38%. When incubated with IAPP at 1:25 and 1:125 ratios, PHEA stars delayed the progression of IAPP toxicity by ~2 h and reduced IAPP-mediated toxicity overall compared to that of the IAPP control. However, with PHEA stars:IAPP 1:5, cells were 94% viable after 24 h, and low levels of cytotoxicity were only observed more than 15 h post incubation. Per their aromatic structures,31 PHEA arms successfully mitigated IAPP cytotoxicity at all concentrations tested (Figure S2). The cytoprotective capacity of PHEA stars at a 1:5 ratio to IAPP was also seen ex vivo, where mouse islets treated with IAPP in the presence of the highest concentration of PHEA stars (~8% relative cell death) demonstrated a significant reduction in toxicity compared to that of IAPP alone (64%) after 48 h treatment.

The near-complete mitigation in IAPP-mediated toxicity observed both in vitro and ex vivo when IAPP was incubated with PHEA stars correlated with stelliform amyloid formation at the PHEA:IAPP ratio of 1:5, as observed in Figure 4A. Likely the key to the cytoprotective nature of stelliform amyloids lies first in their mechanism of formation and additionally in terms of the structure itself. First, oligomeric and low-order protofibrillar species formed as intermediates during IAPP amyloidogenesis are widely considered responsible for the majority of IAPP-mediated cytotoxicity3 with far less toxicity attributed to amyloid fibrils.28 Seeding of IAPP oligomers to the plasma membrane of pancreatic β-cells resulted in destabilization of the lipid membrane47 and cell death through “lipid stripping.”48 Concordantly, rapid sequestering of toxic low-order IAPP species through the formation of stelliform amyloids mediated by PHEA stars would reduce the local population of toxic species around the cell membrane.

Lastly, it has also been purported that the cytotoxicity of amyloid fibrils is mediated through partitioning of the hydrophobic, stiff fibrils into the cell membrane, leading to disruption of the membrane and production of radical oxygen species.17,49,50 The structure of stelliform amyloids, with a compact core and vastly reduced persistence and contour

Figure 5. Protective effect of PHEA stars against IAPP-mediated cytotoxicity in pancreatic beta cells and islets. (A) In vitro cytotoxicity of fibrillating IAPP and mature IAPP amyloids in the presence and absence of PHEA in βTC6 cells over 24 h. Error = SEM (n = 3). (B) Ex vivo cytotoxicity of fibrillating IAPP and IAPP:PHEA at a 1:5 molar ratio in mouse islets after 48 h incubation. Flow cytometry data is representative of n = 5 experiments summarized in the graph. Error = SEM. ****p < 0.0001, one-way ANOVA with Tukey’s correction.
lengths of radiating fibrils, would be unable to effectively partition into the lipid bilayer\(^1\) and would also readily form a protein corona\(^2\)\(^\text{,}^3\) within the extracellular milieu through electrostatic and hydrophobic interactions, further limiting any amyloid contact with cellular membranes. Importantly, we observed complete protection from IAPP-induced cell death in primary mouse islets treated with PHEA stars at a 1:5 molar ratio.

**In Silico Study of the PHEA Polymers and Their Effects on IAPP Aggregation.** To complement the experimental findings, we examined the structural properties of model PHEA polymers by all-atom DMD simulations\(^4\) (see Experimental Methods, ESI). We first studied 2-arm PHEA polymers (Figure S4A) with different degrees of polymerization (DP) and computed their corresponding radius of gyration (\(R_g\)) values (Figure S4D). The data revealed an approximately linear dependence of \(R_g\) on DP (up to \(\sim 40\), Figure S4D), suggesting that the PHEA stars were rather rigid. The autocorrelation analysis of the polymer dynamics in simulations resulted in an estimated Kuhn length of \(\sim 36\) repeats (Figure 6).
S4E), which confirmed the rigidity of the PHEA stars. To evaluate the structure and dynamics of PHEA stars, we studied an 8-arm PHEA model with molecular compositions resembling the experimental data (Figure S4C) in all-atom DMD simulations. A rapid equilibration in terms of $R_e$ and ellipticity was observed (Figure S4F). The average $R_e$ of the 8-arm polymer was $\sim 4.5$ nm, consistent with the experimentally measured hydrodynamic radii (Figure 1C). The high ellipticity value (close to 1) suggests that the 8-arm PHEA adopted a nonspherical conformation as illustrated by a typical snapshot 3D structure (e.g., the inset of Figure S4F). Because of the high rigidity, the interactions between different arms were found minimal beyond the covalent cross-links. Therefore, the all-atom DMD simulations revealed a different morphology of the PHEA stars from that of PAMAM dendrimers, which feature a micellar structure with a porous interior for encapsulating small molecules and IAPP peptides.

To provide molecular insight into IAPP-PHEA binding and its effect on IAPP self-association, we performed DMD simulations on two sets of molecular systems with one containing six IAPP peptides along with a 6-arm PHEA polymer (Figure S4B) and another of six peptides alone as the control (see Experimental Methods in the ESI). We first monitored the size of the largest IAPP aggregates as a function of the simulation time and noted that the presence of PHEA indeed accelerated the self-association clusters in silico (Figure SSA). On the basis of the last 25 ns of the simulations where the largest IAPP aggregates were formed, the binding probability of each IAPP residue with PHEA indicated that both polar and nonpolar residues of IAPP could bind PHEA, though hydrophobic and aromatic residues showed a slightly higher binding propensity (Figure SSB). As a result, the generally nonspecific attraction between IAPP and PHEA led to the accumulation of peptides on PHEA arms, and the increased local peptide concentration accelerated the aggregation of IAPP consistently with a previous coarse-grained computational study. We also examined the secondary structure of IAPPs and their binding with PHEA along the simulation trajectories (e.g., one of the independent simulations shown in Figure S6), where a general trend of correlation between IAPP-PHEA binding and $\beta$-sheet formation in IAPP aggregates was evident. Comparison-average secondary structure contents of the last 25 ns between simulations with and without PHEA (Figure SSC and D) suggest that the PHEA binding did not significantly affect the structures of the aggregates other than accelerated IAPP self-association (Figure S4A).

Next, the kinetics of $\beta$-sheet formation were analyzed for simulations of IAPPs with and without PHEA. The total number of IAPP residues in the $\beta$-sheet conformation followed sigmoidal-like kinetics (i.e., a lag phase followed by rapid growth/elongation and saturation as in Figure S7A) resembling experimentally observed aggregation kinetics. As expected for a nucleation-dependent process, each of the independent simulations rendered different lag times and elongation rates (see the fitting analysis in the Experimental Methods). The presence of PHEA significantly reduced the aggregation lag times (Figure S7B) and broadened the distribution of the elongation rates (Figure S7C), further suggesting that PHEA binding accelerated the nucleation of $\beta$-sheet aggregates and induced heterogeneity in $\beta$-sheet elongation, respectively. Additionally, the potential of mean force (PMF; i.e., the effective aggregate free energy landscape in Figure 6) was computed with respect to the size of IAPP oligomers, $n_{\text{oligomer}}$, and the degree of IAPP fibrillation, $Q_{\text{fibrillation}}$ (for details, see Experimental Methods in the ESI). Two major basins, one corresponding to IAPP monomers and oligomers with little $\beta$-sheets (e.g., highlighted as $a,b$ in Figure 6A and $\alpha,\beta$ in Figure 6B) and the other denoting IAPP aggregates with high amounts of $\beta$-sheets (e.g., $c,d$ in Figure 6A and $\gamma,\delta$ in Figure 6B), could be observed in both IAPP aggregation free energy landscapes with and without PHEA stars. The saddles connecting the two basins corresponded to the aggregation pathways and intermediates (e.g., $a,d$ in Figure 6A and $\gamma,\delta$ in Figure 6B). The presence of the PHEA star rendered the non-$\beta$-sheet basin shallower and the saddle broader (i.e., more pathways/routes toward final $\beta$-rich aggregates in Figure 6B), which accounted for the reduced aggregation lag times (Figure S7B) and heterogeneity in $\beta$-sheet elongation rates (Figure S7C).

### CONCLUSIONS

Inspired by the mechanism of Pmel17 amyloidogenesis, we have developed and established that a polymeric star nanoparticle, PHEA, is capable of mitigating IAPP-mediated toxicity both in vitro and ex vivo through PHEA-mediated promotion of IAPP aggregation and formation of a unique “stelliform amyloid” morphology. Unlike the porous PAMAM-OH dendrimers, which inhibited both IAPP aggregation and toxicity through peptide sequestration, the possession of high rigidity, long arm length, and rich aromatic moieties of PHEA stars facilitated rapid deposition and fibrillation of IAPP monomers into amyloid fibrils. Subsequently, this amyloid structure elicited significantly reduced toxicity in a pancreatic $\beta$-cell line and in mouse islets when compared to the long, semiflexible fibrils typically formed by IAPP. In light of the observation that both PHEA arms and stars elevated IAPP aggregation while mitigating IAPP toxicity, whereas dendrimers and hyperbranched polymers have been predominantly shown in the literature to inhibit amyloid protein aggregation, it is plausible to attribute the observed phenomenon chiefly to the structure and physicochemical properties of the PHEA. Nonetheless, the shape/morphology of the PHEA stars was likely a contributing factor in the IAPP-PHEA interaction, as reflected by the difference in IAPP fibril stiffness associated with PHEA stars versus arms. Shortening of the oligomer lifetime through amyloid aggregation promotion represents a potential strategy to be explored within the larger context of amyloid research, although implementation of such a strategy in vivo is likely a contributing factor in the IAPP-PHEA interaction, as reflected by the difference in IAPP fibril stiffness associated with PHEA stars versus arms. Shortening of the oligomer lifetime through amyloid aggregation promotion represents a potential strategy to be explored within the larger context of amyloid research, although implementation of such a strategy in vivo is likely a contributing factor in the IAPP-PHEA interaction, as reflected by the difference in IAPP fibril stiffness associated with PHEA stars versus arms.

### ASSOCIATED CONTENT

**Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.bioMac.7b01301.

Details of star polymer characterization, PHEA arms analysis, CD spectroscopy, and DMD simulations (PDF)

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