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1	CS ₂ increasing CH ₄ -derived carbon emissions and active
2	microbial diversity in lake sediments
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15 Abstract

Lakes are important methane (CH₄) sources to the atmosphere, especially eutrophic 16 17 lakes with cyanobacterial blooms accompanied by volatile sulfur compound (VSC) emissions. CH₄ oxidation is a key strategy to mitigate CH₄ emission from lakes. In this 18 study, we characterized the fate of CH4-derived carbon and active microbial 19 communities in lake sediments with CS₂ used as a typical VSC, based on the 20 21 investigation of CH₄ and VSC fluxes from Meiliang Bay in Lake Taihu. Stable isotope probing microcosm incubation showed that the efficiency of CH₄-derived carbon 22 23 incorporated into organic matter was 21.1% in the sediment with CS₂ existence, which was lower than that without CS₂ (27.3%). $SO_4^{2-}S$ was the main product of CS₂ 24 oxidation under aerobic condition, accounting for 59.3-62.7% of the input CS₂-S. CS₂ 25 and CH₄ coexistence led to a decrease of methanotroph and methylotroph abundances 26 and stimulated the production of extracellular polymeric substances. CS₂ and its 27 metabolites including total sulfur, SO_4^{2-} and acid volatile sulfur acted as the main 28 drivers influencing the active microbial community structure in the sediments. 29 Compared with α -proteobacteria methanotrophs, γ -proteobacteria methanotrophs 30 Methylomicrobium, Methylomonas, Crenothrix and Methylosarcina were more 31 dominant in the sediments. CH₄-derived carbon mainly flowed into methylotrophs in 32 the first stage. With CH₄ consumption, more CH₄-derived carbon flowed into non-33 methylotrophs. CS₂ could prompt more CH₄-derived carbon flowing into non-34 methanotrophs and non-methylotrophs, such as sulfur-metabolizing bacteria. These 35 findings can help elucidate the influence of VSCs on microorganisms and provide 36 insights to carbon fluxes from eutrophic lake systems. 37

38 Keywords: sulfur-containing compounds; methanotrophs; methane oxidation; CS₂
 39 oxidation rate; sulfur-metabolizing microorganisms

40

41 **1. Introduction**

Methane (CH₄) is an important greenhouse gas with a global warming potential 42 \sim 34 times greater than that of carbon dioxide (CO₂) on a molar basis (Ciais et al., 2013). 43 CH₄ emission includes natural and anthropogenic sources, of which natural sources 44 account for approximately 40% of the global CH₄ emission (Saunois et al., 2020). Lakes 45 are important CH₄ sources to the atmosphere due to gradually increasing emissions with 46 47 global climate change (Guo et al., 2020). Recently, severe eutrophication has led to an increase of cyanobacterial blooms in lakes (Yang et al., 2021), resulting in a pronounced 48 49 increase in CH₄ emission from lakes mainly due to the increase of degradation of dead algae (Bartosiewicz et al., 2021; Zhang et al., 2021). 50

CH₄ produced from organic matter in sediments can be partially oxidized in lake 51 systems when it is emitted to the atmosphere. Methanotrophs are a unique group of 52 Gram-negative bacteria capacity of utilizing -CH₄ as sole carbon and energy source 53 (Semrau et al., 2010). Aerobic methanotrophs mainly belong to Proteobacteria and can 54 55 be classically divided into two types: γ -proteobacteria and α -proteobacteria methanotrophs, based on phylogeny and morphological characteristics (Semrau et al., 56 2010). The γ -proteobacteria methanotrophs belong to the family Methylococcaceae, 57 while α-*proteobacteria* methanotrophs affiliated with the families 58 are Methylocystaceae and Beijerinckiaceae (Bowman, 2006; Semrau et al., 2010). In 59 addition, Verrucomicrobia methanotrophs have been found in extreme environments, 60 extending the diversity of aerobic methanotrophs (Dunfield et al., 2007; Sharp et al., 61 2014). CH₄ is first oxidized to methanol by methane monooxygenase (MMO), 62 including particulate methane monooxygenase (pMMO) encoded by the pmoA gene 63 and soluble methane monooxygenase (sMMO) encoded by the *mmoX* gene (Semrau et 64 al., 2010). Most methanotrophs possess the *pmoA* gene, except *Methylocella* (Dedysh 65

et al., 2000) and *Methyloferula* (Vorobev et al., 2011), while the *mmoX* gene can be
found in some methanotrophs, such as *Methylosinus trichosporium* OB3b and *Methylocella silvestris* BL2 (Semrau et al., 2018). Thus, the *pmoA* gene has been widely
used as a biomarker to identify aerobic methanotrophs.

Volatile sulfur compounds (VSCs) widely exist in terrestrial aquatic systems and 70 are mainly produced from the degradation of organic matter in sediments, which is 71 72 usually accompanied with CH₄ emission (Liu et al., 2019). The VSCs are mainly in the form of hydrogen sulfide (H_2S), methanethiol (CH_3SH), carbon disulfide (CS_2), 73 74 dimethyl sulfide ((CH₃)₂S), and dimethyl disulfide ((CH₃)₂S₂) (Bentley and Chasteen, 2004; Wu et al., 2010). H₂S and CH₃SH can be detected throughout the annual cycle in 75 the water of Lake Taihu, China, ranging from 0.6-882.4 and 0.9-156.3 nmol L⁻¹, 76 respectively (Liu et al., 2019). (CH₃)₂S level is extremely high (up to 6000-7000 nmol 77 L^{-1}) in May and July relative to other seasons when algal blooms form (Reese and 78 Anderson, 2009). CS₂ is found to be the most abundant sulfur compound with a 79 concentration of up to 60000 nmol L⁻¹ in redox transition layer and water interface 80 closed to sediment in Lake Cadagno, Switzerland (Fritz and Bachofen, 2000). In the 81 vertical profile, VSC concentrations such as CH₃SH are shown to have depth gradients, 82 with the maximum CH₃SH concentration just above the sediment (42 nmol L^{-1}) and 83 very low levels (1.1-7.8 nmol L^{-1}) at the water surface (Lomans et al., 1997). In 84 terrestrial aquatic systems, the VSC concentrations change with quality of terrestrial 85 water, oxygen concentration, temperature, light intensity and eutrophication level 86 (Watson and Jüttner, 2017). 87

VSCs such as H_2S , $(CH_3)_2S$ and CH_3SH have been reported to have inhibitory effects on CH_4 oxidation due to the potential toxicity of sulfur-containing compounds and their metabolic intermediates (Cooper and Brown, 2008; Lee et al., 2011; Long et

al., 2013; Zhang et al., 2016). Lee et al. (2011) found that CH₄ oxidation could be 91 inhibited in the presence of H₂S, CS₂, (CH₃)₂S or CH₃SH. The growth of methanotrophs 92 can be inhibited at a H₂S concentration of 0.01%, while CH₄ oxidation capacity may be 93 inhibited at a H₂S concentration of over 0.05% (Caceres et al., 2014; Zhang et al., 2016). 94 Long et al. (2013) found competitive inhibition between H₂S and CH₄ when CH₄ was 95 at a low concentration of 5%. Börjesson (2001) reported a negative correlation between 96 97 methanotrophic activity and CS₂ concentration. Saari and Martikainen (2003) found a negligible effect on CH₄ oxidation at a (CH₃)₂S concentration of 0.64 µmol in soil-water 98 99 slurry, while complete inhibition was observed at a DMS concentration of 32 µmol. Yu et al. (2009) reported that H₂S in coal mine gas did not have an obvious influence on 100 MMO activity. However, Kim et al. (2013) found that (CH₃)₂S had a positive influence 101 on methanotrophs, favoring the growth of Methylosarcina. Eutrophic shallow 102 freshwater lakes are ecosystems in which CH₄ and sulfur-containing compounds 103 coexist, especially when algal blooms form. However, little is known about the effects 104 of sulfur-containing compounds on the activity and communities of methanotrophs in 105 freshwater lakes. 106

Methanotrophs can not only oxidize CH₄, but also play a key role in metabolizing 107 sulfur-containing compounds. Compared with pMMO, sMMO has activity with a broad 108 range of substrates (Baker et al., 2001). Sorokin et al. (2000) identified a new type I 109 110 methanotrophic strain (Methylomicrobium) that could oxidize CS₂ at a high pH of 9-10, suggesting that CS_2 might be an available substrate for MMO. Moreover, this strain 111 also has the capacity for (CH₃)₂S oxidation. Additionally, methanotrophs can excrete 112 intermediates from CH₄ oxidation, such as methanol, formaldehyde and formate, which 113 can be utilized by non-methanotrophs such as sulfur-metabolizing microorganisms as 114 carbon sources (Xin et al., 2007; Tavormina et al., 2017). Environmental factors such 115

as nitrate or low O₂ level can promote flux of CH₄-derived carbon from CH₄ oxidation
to other non-methanotrophs (Wei et al., 2016; He et al., 2020; van Grinsven et al., 2021).
However, a clear understanding of the effects of VSCs on the metabolism of
methanotrophs and the feedback on carbon cycling in lakes is still lacking.

The objective of this work was to characterize the fate of CH₄-derived carbon and 120 active methanotrophic communities in lake sediments by applying DNA-stable isotope 121 122 probing (DNA-SIP) technique. CS₂ was chosen as a typical VSC owing to the highdetected CS₂ flux from Meiliang Bay in Lake Taihu in this study. CH₄ oxidation 123 124 potential was determined during the SIP incubation. Organic matter characteristics including the contents of organic matter, organic ¹³C and extracellular polymeric 125 substances (EPS) were analyzed during the experiment. The microbial abundance and 126 community in the sediments were analyzed by quantitative PCR (q-PCR) and MiSeq 127 sequencing of the 16S rRNA gene. We hypothesized that different groups of 128 methanotroph would have different CS₂ tolerances which would affect CH₄ oxidation 129 rates, fate of CH₄-derived carbon and microbial communities in lake sediments. A 130 deeper investigation of the effects of CS₂ on the fate of CH₄-derived carbon and 131 communities of methanotrophs can help elucidate the response of methanotrophs to 132 VSCs and establish models based on environmental factors to estimate CH₄ emission 133 from shallow eutrophic lakes. 134

135

136 2. Materials and methods

137 *2.1. Study site and sampling*

Lake Taihu is a typical large shallow lake with an average water depth of 1.9 m and a surface area of 2338 km² located in the Yangtze River Delta Plain, China (Qin et al., 2007). The study sites were located in Meiliang Bay, which is in the northwestern area of Lake Taihu. Due to the predominance of southeasterly winds and nutrient input,
Meiliang Bay suffers from severe levels of eutrophication, frequently leading to intense
algal blooms and black-colored water (Duan et al., 2015; Wu et al., 2018).

CH₄ and VSC fluxes were detected in four sites (120°9'31"-120°10'42", 31°28'10"-144 31°28'35") of Meiliang Bay using the floating chamber technique (St Louis et al., 2000; 145 Gonzalez-Valencia et al., 2014). The static chamber was cylindrical with an open 146 surface of 0.4 m² and a headspace volume of 6.5×10^{-2} m³. Each static chamber was 147 equipped with a sampling port. Gas samples were collected from the static chambers 148 149 every 20 min within a 1-h period in triplicate and stored in pre-vacuumed 20-mL vials for detecting CH₄ and VSC concentrations within 24 h. Three static chambers with 1-150 m interval were used to measure gas fluxes in each site. 151

The sediment cores were collected from four sites in Meiliang Bay with a gravity 152 corer equipped with Perspex tubes (inner diameter, 10 cm; length, 35 cm) in November 153 2019 with the average temperature, pH, dissolved oxygen (DO) and chlorophyll a of 154 18°C, 7.7, 6.41 mg L⁻¹, and 2.66 µg L⁻¹, respectively. A total of eight sediment cores 155 were obtained with two cores sampled at each site. Overlying water was retained in the 156 core tubes to prevent the turbulence and oxidation of surface sediments. The sediment 157 cores were immediately transported to the laboratory, carefully extruded and sliced into 158 the top 2 cm of sediment after removing the overlying water. Then, the surface 2 cm 159 sediment was homogenized and placed into plastic freezer bags. Approximately 1 kg 160 sediment subsamples were stored at 4 °C for SIP incubation and analysis of 161 physicochemical properties, including organic matter (OM), sulfur compounds (SO42--162 S, acid volatile sulfide (AVS), total sulfur (TS)) and EPS including extracellular 163 polysaccharide (ECPS) and extracellular protein (ECP). The remaining subsamples 164 were stored at -80 °C for further molecular analysis. 165

167 2.2. SIP microcosms and CH₄ oxidation potential

Approximately 5 g (wet weight) of homogenized sediment was placed into 200 mL 168 sterile serum vials sealed with a butyl rubber stopper. After extracting 4 mL gas, ¹³CH₄ 169 (99 atom % ¹³C, Sigma-Aldrich) was injected into the serum vials to a final 170 concentration of 2% in the headspace (v/v). Two groups were set up: 1) only ${}^{13}CH_4$ was 171 injected into the serum vials, labeled "M"; 2) in addition to ¹³CH₄, CS₂ was injected to 172 the initial concentration of 350 mg m⁻³ in the headspace, labeled "MS". Controls were 173 amended with ¹²CH₄ (99.9% purity). Each ¹³CH₄ or CH₄ group was performed in 15 174 replicates. Autoclaved sediment with the addition of NaN₃ $(0.13 \text{ g kg}_{dry \text{ wet}}^{-1})$ was used 175 to detect abiotic CH₄ and CS₂ loss in triplicate. The microcosms were flushed with air 176 when more than 90% of CH₄ was consumed. Then CH₄ and CS₂ were added to obtain 177 the initial concentrations as described above. There were a total of four gas exchanges 178 during the whole incubation period. CS₂ was directly injected into the microcosms to 179 achieve an initial concentration of 350 mg m⁻³ when the CS₂ concentration was below 180 10 mg m⁻³ and when the concentrations of CH₄ and O₂ were higher than 5% in the 181 headspace of the serum vials. CS₂ was injected twice, 8 and 29 times when the SIP 182 samples were harvested at the CH₄ consumption of 20, 50 and 100 µmol CH₄ g_{wet weight} 183 ¹, respectively. Five serum vials were randomly taken for the destructive sampling of 184 sediment for physicochemical properties, ¹³C amount and molecular analysis when the 185 CH₄ consumption reached 20, 50 and 100 µmol CH₄ gwet weight⁻¹. Gas samples were 186 extracted from the headspace of the serum vials to measure CH₄ and CS₂ concentrations 187 each day. All serum vials were incubated in the dark at 30°C to simulate the high 188 temperature of 32-34 °C detected in overlying water in summer. The experiment lasted 189 35 days. 190

191

192 *2.3. Analytical methods*

CH₄ concentration was detected by using a gas chromatograph with a thermal 193 conductivity detector, which was equipped with a GDX-104 (60/80 mesh) paced 194 column of 2 m length (He et al., 2018b). The temperatures of oven, injector and detector 195 were 60, 80 and 100°C, respectively. N₂ was used as carrier gas at a flow rate of 20 mL 196 197 min⁻¹. VSC concentrations were detected by using the HC-3 trace sulfur analyzer equipped with a flame photometric detector and a polytetrafluoroethylene column as 198 199 described previously (Chen et al., 2017). The temperatures of the oven, injector and detector were 80, 90 °C and room temperature, respectively. The CH4 oxidation 200 potential was estimated from a zero-order decrease in the headspace of serum vials 201 202 (Wang et al., 2011). The gas fluxes including CH₄ and VSCs were calculated as described by Gonzalez-Valencia et al. (2014). 203

The OM content of sediment was determined as described previously (Bao, 2000). The EPS content of sediment was extracted and detected as described by Wei et al. (2015). The contents of sulfur-containing compounds, including TS, AVS, and $SO_4^{2^-}$, were determined as described previously (He et al., 2018a). The total ¹³C, ¹³C of organic C and ¹³C of CO₂-C in the sediment samples were detected using a Elementar vario MICRO cube elemental analyser coupled to GV Isoprime 100 isotope ratio mass spectrometer (GV Instruments, U.K.) as described by He et al. (2020).

211

212 2.4. DNA extraction and gradient centrifugation

Approximately 0.5 g sediment subsamples were extracted using E.Z.N.A. TM soil DNA extraction kit (Omega Bio-Tek, Inc., Norcross, USA) as described in the manufacturer's instructions. DNA concentration was detected with a NanoDrop 2000

Spectrophotometer (Thermo Fisher Scientific, United States). DNA was subjected to 216 equilibrium density gradient centrifugation using cesium chloride (CsCl). In brief, 217 approximately 3 µg DNA was combined with CsCl solution and loaded into a centrifuge 218 tube (Himac, Koki Holdings Co., Ltd, USA), spun in a p90NT vertical rotor (Beckman 219 Coulter, CA) and ultracentrifuged at 50800 rpm at 20 °C for 60 h. Twenty density 220 gradient fractions were obtained, and the buoyant density (BD) of fractions was 221 222 measured by weighing aliquots. The fractionated DNA was precipitated using polyethylene glycol 6000 (PEG6000) overnight at 4 °C. Then, the pellets were purified 223 224 with 70% ethanol twice and resuspended in 20 µL nuclease-free water after centrifugation at 13000 g for 30 min. Heavy DNA was identified via a quantitative PCR 225 (qPCR) analysis of 16S rRNA gene as described by Leigh et al. (2007) with the primer 226 227 of 338F 5'-ACTCCTACGGGAGGCAGCAG-3'/515R 5'set ATTACCGCGGCTGCTGG-3' (Fierer et al., 2005). After identifying the ¹³C-labeled 228 DNA fractions, the fractions were combined to form the "heavy" fraction (¹³C-labeled 229 DNA) for further molecular analyses. 230

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232 2.5. Q-PCR analyses

Total DNA was used to determine the abundance of 16S rRNA gene (Fierer et al., 233 2005), pmoA, mmoX and mxaF using qPCR analysis. qPCR was performed in triplicate 234 235 with a 15 μ L reaction mixture for each sample including 1 × qPCR mix (GeneCopoeia, USA), 0.4 μ mol L⁻¹ primers, and 1 μ L DNA template. The primers for the *pmoA*, *mmoX*, 236 and mxaF genes were A189F 5'-GGNGACTGGGACTTCTGG-3'/Mb661R 5'-237 CCGGMGCAACGTCYTTACC-3' (Kolb et al., 2003), mmoX206F 5'-ATCGCB 238 5'-ACCCANGGCTCGACYTTGAA-3' AARGAATAYGCSCG-3'/mmoX886R 239 (Hutchens et al., 2004), and mxa1003F 5'-GCGGCACCAACTGGGGCTGGT-240

3'/mxa1561R 5'-GGGCAGCATGAAGGGCTCCC-3', respectively (McDonald et al., 1997). The thermal conditions set for 16S rRNA gene were as follows: initial denaturation at 95 °C for 10 min followed by 40 cycles of 95 °C for 30 s, 57 °C for 30 s, and 72 °C for 30 s. The annealing temperatures of *pmoA*, *mmoX* and *mxaF* were 58 °C, 56 °C, and 55 °C, respectively. The standard curves were performed in a range of 10^{-1} - 10^{-7} ng μ L⁻¹. The amplification efficiencies of these genes were between 90% and 93% with R² values of 0.90-0.99.

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249 2.6. 16S rRNA gene MiSeq sequencing

Total DNA and ¹³C-DNA retrieved from "heavy" fractions were used for PCR 250 amplification. The PCR mixture in a total volume of 50 μ L contained 5 μ L 10 × buffer, 251 252 0.2 mM dNTPs, 0.06 U DNA Polymerase, 0.8 mM 338F/806R primers with barcodes, and 2 µL DNA template. PCR amplification was performed at 95 °C for 3 min, followed 253 by 30 cycles of 95 °C for 30 s, 55 °C for 30 s, and 72 °C for 45 s, with a final extension 254 at 72 °C for 10 min. PCR products were verified with 1.2% (w/v) agarose gels and 255 sequenced using the Illumina MiSeq platform at Shanghai Majorbio Bio-Pharm 256 Technology Co., Ltd. 257

The sequencing data were analyzed on the online Majorbio Cloud platform (www.majorbio.com) for quality filtering, chimera removal, and operational taxonomic unit (OTU) clustering on QIIME (v1.9.1). The paired-end reads were trimmed with Flash (v1.2.11). OTUs were clustered on the basis of a 97% similarity using Uparse (v7.0.1090). Sequencing data was classified with RDP Classifier (v2.11). The sequences obtained in this study have been deposited in the National Omics Data Encyclopedia (NODE) database under accession number OER221345-OER221422.

265

266 *2.7. Data analysis*

Statistical analysis was performed by SPSS Statistics 23.0 based on Student's t-test and analysis of variance (ANOVA). Alpha diversity was calculated in Mothur (v1.30.2) and visualized with the ggplot2 package in R (v4.0.3). Redundancy analysis (RDA) was performed based on the OTU level measured in R (v4.0.3) with the vegan package and visualized with the ggplot2 package.

272

273 **3. Results and discussion**

274 *3.1. CH*⁴ *and VSC fluxes*

CH₄ and VSC fluxes from the Meiliang Bay were detected in four seasons (i.e., 275 March, May, August and November) using floating static chambers. The CH₄ fluxes 276 from the Meiliang Bay ranged from 16.8 to 78.7 mg m⁻² d⁻¹ (Fig. 1). Compared with 277 the other seasons, the CH₄ fluxes were lower in March, due to the lower temperature of 278 14°C which influenced the degradation of organic matter. Similar studies have reported 279 that exponential increases in CH₄ emissions with increasing temperature in freshwater 280 systems (Aben et al., 2017; van Bergen et al., 2019). H₂S and CS₂ were the two detected 281 VSCs with the fluxes of 10.7-65.4 and 14.4-72.7 mg m⁻² d⁻¹, respectively, from the 282 Meiliang Bay. The H₂S fluxes were similar to the CH₄ emissions with the lowest 283 occurring in March. Statistical analysis showed that the H₂S emissions had a strong 284 285 linear correlation with the CH₄ fluxes from the Meiliang Bay. The CS₂ fluxes were low with the average of 22.1 mg m⁻² d⁻¹ in March, May and August. However, the CS₂ flux 286 was in range of 35.9 to 72.7 mg m⁻² d⁻¹ in November, which was significantly higher 287 288 than the other seasons. This might be photochemical CS₂ production from the precursor molecules such as cysteine, cystine and (CH₃)₂S of Cyanobacterial dying and 289 degradation (Modiri Gharehveran and Shah, 2018) and a biological production of 290

291 phytoplankton species (Xie et al., 1999). Further studies such as CS₂ formation need to

be conducted to illustrate the high CS₂ fluxes from the Meiliang Bay in November.

293

294 *3.2. CH*⁴ *oxidation and fate of CH*⁴*-derived carbon*

After exposure to CH₄, the CH₄ oxidation potential (MOP) of sediment increased 295 in the first two days (Fig. 2A). The MOP of sediment reached the peak value (2.8 μ g g⁻ 296 ¹ h⁻¹) in the M group on day 2, which was significantly higher than that in the MS group, 297 likely due to the toxicity of CS₂ to microorganisms (Cooper and Brown, 2008). 298 299 Börjesson et al. (2001) also found that high concentrations of CS₂ addition could lead to limited methanotrophic activity. Thereafter, the MOP of sediment was not obviously 300 different between the MS and M groups and fluctuated from 2.5-2.8 µg g⁻¹ h⁻¹, 301 indicating that a CS₂ concentration of 350 mg m⁻³ had no significant effect on MOP in 302 a longer term. 303

During CH₄ oxidation, CH₄-derived carbon was mainly converted into CO₂-C, 304 biomass-C and organic-C such as EPS, formate, acetate and lactate (He et al., 2020). 305 The OM content was 26.3 g kg⁻¹ in the original sediment (Fig. 2B). After exposure to 306 CH₄, the OM content of sediment increased with continuous CH₄ consumption and 307 reached 31.2 and 32.8 g kg⁻¹ in the MS and M groups, respectively, when the CH₄ 308 consumption was 100 µmol g_{wet weight}⁻¹, owing to the secretion of organic matter such as 309 acetate and oxalate during CH₄ oxidation (Kalyuzhnaya et al., 2013; Ma et al., 2021). 310 Compared with the M group, the increase of the OM content was slightly less in the 311 MS group, however, no significant difference in the OM content of sediment was found 312 between the M and MS groups. This indicated that CS₂ addition could influence the 313 distribution of CH₄-derived carbon and reduce organic carbon deposited into the 314 sediment. 315

Carbon conversion efficiency (CCE) refers to the amount of organic carbon (OC) 316 produced from CH₄ oxidation, which can be determined by a regression analysis of the 317 amount of OC produced from CH₄ oxidation with CH₄-C consumption (Trimmer et al., 318 2015). In this study, a strong linear relationship was found between the organic ^{13}C 319 amount and ¹³CH₄ consumption with the correlation coefficients (R) of 0.997 and 0.999 320 321 for the M and MS groups, respectively (Fig. 2C). Compared to the CCE of 21.1% in 322 the MS group, the CCE was higher (27.3%) in the M group, indicating that more CH₄derived carbon was converted into organic carbon and deposited into the sediments in 323 324 the M group than in the MS group. This result was in accordance with the OM content (Fig. 2B) of sediment, which might be because CS₂ might prompt more CH₄-derived 325 carbon used for the growth and metabolism of CS2-degrading microorganisms. 326 Previous studies have also shown that CH₄-derived carbon can be served as nutrients to 327 higher trophic levels and be influenced by various environmental factors such as O₂ 328 concentrations, nitrogen sources, and the existence of toluene (Medvedeff and Hershey, 329 2013; He et al., 2015; He et al., 2020). 330

The efficiency of CH₄-derived carbon conversion into total carbon (TC) of 331 sediment (CE_{TC}) was 31.5% when the CH₄ consumption was 20 µmol g_{wet weight}⁻¹, 332 increased with CH₄ consumption and reached 41.8% in the M group when the CH₄ 333 consumption was 50 µmol gwet weight⁻¹ (Fig. 2D). Relative to the M group, the CE_{TC} level 334 was significantly lower in the range of 24.7%-27.5% in the MS group. The efficiency 335 of CH₄-derived carbon conversion into CO₂-C (CE_{CO2-C}) was 52.5%-57.1% and 62.5%-336 64.7% in the M and MS groups, respectively (Fig. 2E). The CE_{CO2-C} levels in the MS 337 group were significantly higher than in the M group. This indicated that CS₂ addition 338 could promote more CH₄-C converted into CO₂-C and decrease the TC content of the 339 sediment. 340

341

342 *3.3. Secretion of EPS*

EPS is a complex mixture formed by biochemical processes from high molecular 343 weight organic compounds, including cell exudates, autolysis and cell surface material 344 shedding (Costa et al., 2018). EPS is comprised of carbohydrates, proteins, nucleic 345 acids, humic acids and other substances, of which ECPS and ECP account for 75-89% 346 347 (McSwain et al., 2005; Ni et al., 2009). The ECPS and ECP contents were about 71.8 and 227.3 mg kg⁻¹, respectively, in the original sediment (Fig. 3). After exposure to CH₄ 348 and CS₂, the ECPS content increased and reached 215.0 and 155.3 mg kg⁻¹ in the MS 349 and M groups, respectively, when the CH₄ consumption was 100 μ mol g_{wet weight}⁻¹. The 350 ECP content was 2-3 times of the ECPS content measured in the experimental 351 sediments, which might be attributed to the fact that the availability of substrates could 352 trigger different biosynthesis pathways of EPS. Bozal et al. (1994) reported that 353 Alteromonas strain secreted more protein when growing on a mineral medium, while 354 higher carbohydrate levels of EPS were obtained in a complex medium. Compared with 355 the M group, the ECPS and ECP contents were significantly higher in the MS group 356 when the CH₄ consumption was 100 μ mol g_{wet weight}⁻¹. This may be attributed to the 357 toxicity of CS₂ and its intermediates to microorganisms, which might prompt EPS 358 secretion for survival and defensive functions (Marvasi et al., 2010). Li (2020) also 359 found that Methylomonas koyamae could secrect more EPS into the culture under the 360 CH₃SH stress of 200-1000 mg m⁻³ than the group without CH₃SH. Moreover, CH₃SH 361 and CS₂ could be converted between each other (Li, 2020). Previous studies have also 362 reported that microorganisms could secrete EPS to improve resistance under harsh 363 environments in the presence of heavy metals, chemicals and antibiotics (Fang et al., 364 2002; Coburn et al., 2016). 365

367 *3.4. Variation of sulfur-containing compound contents*

To assess CS₂ metabolism, sulfur-containing compounds, including SO₄²⁻-S, AVS 368 and TS, were determined in the experimental sediments. The TS content was 69.3 mg 369 kg⁻¹ in the original sediment. With CS₂ addition, the TS content of sediment increased 370 and reached the maximum of 287.1 mg kg⁻¹ when the CH₄ consumption was 100 µmol 371 gwet weight⁻¹, while the TS content in the M group almost stayed constant (Fig. 4A). The 372 TS increment in the sediments accounted for 79.8%, 96.7% and 88.0% of the CS₂-S 373 input when the CH₄ consumption was 20, 50 and 100 µmol gwet weight⁻¹, respectively. 374 This indicated that the increase of TS content in the sediments was attributed to the 375 CS₂-S input. Thus, we inferred that CS₂-S could be oxidized or metabolized by 376 microorganisms in the sediment. Hartikainen et al. (2001) also found Thiobacillus could 377 remove CS_2 and produce SO_4^{2-} in a peat biofilter. AVS is a product of sulfate reduction 378 such as the biodegradation of OM with SO42- serving as an electron acceptor (Richard 379 and Morse, 2005). The AVS content increased with incubation time and reached 12.6 380 mg kg⁻¹ in the MS group when the CH₄ consumption was 50 µmol g_{wet weight}⁻¹ (Fig. 4B), 381 which was significantly higher than the others (P < 0.05). The AVS content decreased 382 slightly when the CH₄ consumption was 100 µmol gwet weight⁻¹ in the MS group, likely 383 due to increased AVS oxidation into SO_4^{2-} . 384

The SO₄²⁻-S content was 57.8 mg kg⁻¹ in the original sediment (Fig. 4C). With CS₂ addition, the SO₄²⁻-S content of sediment increased with incubation time and reached 67.3, 97.5 and 223.4 mg kg⁻¹ when the CH₄ consumptions were 20, 50 and 100 μ mol g_{wet weight}⁻¹, respectively, in the MS group, which was in accordance with trends of TS content in sediment (Fig. 4A). The percent of SO₄²⁻-S increment relative to CS₂-S were 59.3%, 61.1%, and 62.7% when the CH₄ consumptions were 20, 50 and 100 μ mol g_{wet} weight⁻¹, respectively, in the MS group (Fig. 4D). This suggested that CS₂-S was mainly oxidized to SO_4^{2-} as the final product in the experimental sediments, owing to the fact that CS₂ was the only S source in MS group. Alcantara et al. (1999) also found that the end-product of microbial CS₂ oxidation was SO_4^{2-} accompanied by the intermediates such as H₂S, COS and elemental S. Additionally, sulfur-containing compounds such as H₂S, CH₃SH and (CH₃)₂S could also be mainly oxidized to SO_4^{2-} under aerobic condition (Fang et al., 2015; Wang et al., 2021).

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399 *3.5. Abundance of bacteria, methanotrophs and methylotrophs*

The abundance of bacterial 16S rRNA gene was $3.88 \times 10^{10} \pm 1.38 \times 10^9$ copies g⁻ 400 ¹ in the original sediment (Fig. 5A). After exposure to CS_2 and CH_4 , the abundance of 401 bacterial 16S rRNA gene decreased to $2.34 \times 10^{10} \pm 3.55 \times 10^{8}$ and $1.74 \times 10^{10} \pm 3.06 \times$ 402 10^8 copies g⁻¹ in the M and MS groups when the CH₄ consumption was 20 µmol g_{wet} 403 weight⁻¹, respectively, likely due to only CH₄ or CH₄ and CS₂ being available as carbon 404 sources for the sediment microbial community. Compared with the M group, the 405 abundance of bacterial 16S rRNA gene was lower in the MS group during the 406 incubation (P < 0.05), suggesting that CS₂ could strongly inhibit microbial growth 407 owing to its toxicity. It has similarly been found that CS2 can inhibit the growth of 408 bacterial species in the rhizosphere (Hartel and Haines, 1992). 409

The abundances of *pmoA* and *mmoX* encoding pMMO and sMMO were 1.77×10^{8} $\pm 6.94 \times 10^{6}$ and $7.56 \times 10^{7} \pm 8.03 \times 10^{6}$ copies g⁻¹, respectively, in the original sediment. After exposure to CH₄, the abundance of *pmoA* and *mmoX* gradually increased to $2.26 \times 10^{9} \pm 1.05 \times 10^{8}$ and $0.38 \times 10^{9} \pm 0.21 \times 10^{8}$ copies g⁻¹, respectively, in the M group when the CH₄ consumption increased to 100 µmol g_{wet weight}⁻¹ (Fig. 5B). However, in the MS group, the abundances of *pmoA* and *mmoX* reached the maxima of

 $1.65 \times 10^9 \pm 8.28 \times 10^6$ and $0.26 \times 10^9 \pm 0.31 \times 10^8$ copies g⁻¹, respectively, when the 416 CH4 consumption was 50 μ mol g_{wet weight}⁻¹ and decreased slightly at a CH4 consumption 417 of 100 µmol gwet weight⁻¹ (Fig. 5C). This might be attributed to the inhibitory effect of 418 CS_2 and its metabolic intermediates and products such as $\mathrm{H}_2\mathrm{S}$ and $\mathrm{SO_4^{2-}}$ on the 419 methanotrophic growth (Long et al., 2013; Caceres et al., 2014). Compared with the M 420 group, the abundance of *pmoA* and *mmoX* was significantly higher in the MS group 421 when the CH₄ consumption was 50 μ mol g_{wet weight}⁻¹ (P < 0.01), indicating that CS₂ 422 423 could enhance methanotrophic growth to a certain extent. However, the stimulating mechanism of CS₂ for methanotrophic growth was not clear. Further studies such as the 424 effect of CS₂ on gene expression need to be conducted to understand the effect of CS₂ 425 426 on methanotrophic growth and metabolism.

Methanol is an important intermediate product during CH₄ oxidation and can be 427 converted into formaldehyde catalyzed by methanol dehydrogenase (MDH) (Islam et 428 al., 2021). The mxaF gene encoding the large subunit of MDH is considered a 429 functional marker gene for identifying methylotrophs. The mxaF abundance was 430 $1.77 \times 10^8 \pm 1.74 \times 10^7$ copies g⁻¹ in the original sediment, identical to the abundance of 431 the *pmoA* gene. After exposure to CH_4 and CS_2 (Fig. 5D), the *mxaF* abundance 432 increased and reached $7.78 \times 10^8 \pm 1.36 \times 10^7$ copies g⁻¹ in the M group when the CH₄ 433 consumption was 100 μ mol g_{wet weight}⁻¹, while it reached the maximum of 6.97×10⁸ ± 434 3.68×10^7 copies g⁻¹ in the MS group when the CH₄ consumption was 50 µmol g_{wet weight} 435 ¹. This was in accordance with the variation of pmoA and mmoX abundance in the 436 sediment samples. Similar results were obtained by He et al. (2012, 2020), who reported 437 a linear relationship between methanotrophs and methylotrophs during CH₄ oxidation. 438

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440 *3.6. Variation of microbial community structure*

SIP microcosms were applied to track the active microbial community assimilating 441 CH₄ and CH₄-derived carbon. The "heavy" and "light" DNA fractions were separated 442 by CsCl gradient ultracentrifugation and identified according to the relative abundance 443 of the 16S rRNA gene by qPCR analysis (Fig. S1). MiSeq sequencing results showed 444 that a diverse bacterial community was present in total DNA and ¹³C-DNA from the 445 experimental sediments. After exposure to CH4 and CS2, the Chao index of the 446 microbial community in the experimental groups increased relative to the original 447 sediment in the total DNA (Fig. 6A). The Chao index of ¹³C-DNA was significantly 448 higher in the MS group than in the M group when the CH₄ consumption was 50 and 449 100 μ mol g_{wet weight}⁻¹ (Fig. 6B). This indicated that after exposure to CS₂, the richness 450 of the active microbial community increased due to more CH4-derived carbon flowing 451 into the microbial community, such as CS₂-metabolizing microorganisms. 452

Proteobacteria, Nitrospirota, Bacteroidota, Desulfobacterota, Chloroflexi, 453 Acidobacteriota, Verrucomicrobiota, Mvxococota and Bdellovibrionota were the main 454 phyla in the sediment samples, accounting for 81.1%-89.4% of the sequencing reads in 455 the total DNA during the whole experiment (Fig. 6C). After exposure to CH₄ and CS₂, 456 the relative abundances of Proteobacteria, Bacteroidota and Acidobacteriota decreased 457 from 54.1%, 15.6% and 7.5% in the original sediment to 31.2%-33.6%, 10.5%-10.9%, 458 and 4.6%-5.0% in the sediment samples, respectively, when the CH₄ consumption was 459 umol g_{wet weight}⁻¹. However, the relative abundances of *Nitrospirota*, 100 460 Desulfobacterota and Chloroflexi, respectively, increased to 11.2%-12.2%, 6.6%-10.8% 461 and 7.6%-8.8% in the sediment samples, being 2-4 times of the levels of the original 462 samples. The relative abundances of Verrucomicrobiota, Myxococcota and 463 Bdellovibrionota increased slightly after exposure to CH₄ and CS₂. The community 464

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structure in phylum level was in accordance with Chao index that there was significant
difference between the original sample and the others, while no obvious difference was
observed between the MS and M groups.

Proteobacteria was the most dominant phylum in the ¹³C-DNA, accounting for 468 39.3%-71.8% of the sequencing reads during the whole experiment, exceeding that of 469 the total DNA (28.4-35.8%) (Fig. 6D). The relative abundances of Bacteroidota and 470 Chloroflexi were respectively 1.2%-3.3% and 2.3%-5.6% in the ¹³C-DNA of the two 471 groups when the CH₄ consumption was 20 µmol g_{wet weight}⁻¹ and respectively increased 472 to 5.9%-12.1% and 3.3%-11.5% when the CH₄ consumption was 100 μ mol g_{wet weight}⁻¹. 473 This indicated that more CH4-derived carbon flowed into Bacteroidota and Chloroflexi 474 with increasing CH₄ consumption. Acidobacteriota and Myxococcota were also 475 abundant in the ¹³C-DNA of the two groups, and a higher relative abundance was found 476 in the ¹³C-DNA of the MS group when the CH₄ consumption reached 20 µmol gwet weight 477 ¹ than in the others, suggesting that *Acidobacteriota* and *Myxococcota* might be related 478 to CS₂ metabolism. Masuda et al. (2016) aslo found Acidobacteriota dominated in the 479 metabolism of sulfur-compounds in paddy soil. Compared with the total DNA, the 480 relative abundances of Nitrospirota, Desulfobacterota and Verrucomicrobiota were 481 lower in the ¹³C-DNA of the two groups during the whole incubation period, suggesting 482 that Nitrospirota, Desulfobacterota and Verrucomicrobiota took part in assimilating 483 CH₄-derived carbon, but their activities were not pronounced. 484

To illustrate the relationships of the sediment samples among environmental factors including TS, SO_4^{2-} , AVS, ECP, ECPS and OM, RDA was conducted based on the OTUs of the total DNA and ¹³C-DNA (Fig. 7) at a 97% similarity level. After exposure to CH₄ and CS₂, the microbial community structure of the sediment samples changed and differed from that of the original sediment. Among the six environmental

factors, TS, SO₄²⁻ and AVS were the main environmental factors influencing the 490 microbial communities in the total DNA and ¹³C-DNA of the MS group, while OM was 491 the main factor affecting the microbial communities in the M group. The total and active 492 microbial communities in the MS and M groups respectively clustered together, 493 suggesting that CS₂ had an important influence on the microbial community structure 494 of total and active microorganisms in assimilating CH₄-derived carbon. A similar result 495 496 was obtained by Shi et al. (2011) who found that the sulfur addition could increase of the abundance of Thiobacillus to oxidize sulfur. 497

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499 *3.7. Active methanotrophs, methylotrophs and sulfur-metabolizing bacteria*

 α -proteobacteria methanotroph Methylocystis and γ -proteobacteria methanotrophs 500 Methylomicrobium, Methylomonas, Crenothrix and Methylosarcina were the dominant 501 methanotrophs in the sediment samples. The relative abundance of methanotrophs was 502 0.4% in the original sediment. After exposure to CH₄, the relative abundance of 503 methanotrophs increased with CH₄ consumption and reached 7.6% in the total DNA of 504 the M group when the CH₄ consumption was 100 µmol gwet weight⁻¹ (Fig. 8A). The 505 relative abundance of methanotrophs also increased in the total DNA of the MS group, 506 and reached a maximum of 6.0% at the CH₄ consumption of 50 µmol g_{wet weight}⁻¹. In the 507 ¹³C-DNA of the M group, the relative abundance of methanotrophs was highest (17.7%) 508 at the CH₄ consumption of 20 µmol gwet weight⁻¹, and decreased with CH₄ consumption, 509 indicating the growth of non-methanotrophs derived from CH₄ oxidation, such as the 510 cross-feeding observed between methanotrophs and methanol-utilizing bacteria 511 (Krause et al., 2017). However, in the ¹³C-DNA of the MS group, the relative abundance 512 of methanotrophs increased and reached the maximum of 27.9% at the CH4 513 consumption of 50 µmol gwet weight⁻¹, indicating CS₂ addition could increase the relative 514

abundance of methanotrophs in the initial stage. This was in accordance with the 515 abundance of *pmoA* and *mmoX* determined by q-PCR analysis. The relative abundance 516 of methanotrophs decreased to 9.7% at the CH₄ consumption of 100 μ mol g_{wet weight}⁻¹, 517 which might be attributed to the toxicity of CS_2 and the intermediates (Caceres et al., 518 2014). The relative abundances of *Crenothrix* and *Methylosarcina* were significantly 519 higher in the M group than in the MS group due to the sensitivity to environmental 520 521 stress (Cruaud et al., 2019). Conversely, the relative abundance of Methylocystis reached higher proportions in the total and ¹³C-DNA of the MS group than those of the 522 523 M group owing to the existence of sMMO, which showed broad substrate availability (Lee et al., 2006). 524

Methylophilus, *Methylotenera*, *Methyloversatilis*, Hyphomicrobium 525 and Methylobacillus were the main methylotrophs in the total DNA of sediments (Fig. 8B). 526 The relative abundance of methylotrophs was less than 0.2% in the original sediment. 527 After exposure to CH₄, the relative abundance of methylotrophs increased and reached 528 the maximum of 5.1% in the total DNA of the M group at the CH₄ consumption of 50 529 μ mol g_{wet weight}⁻¹, while it increased with incubation time and reached 4.1% in the total 530 DNA of the MS group at the CH₄ consumption of 100 µmol g_{wet weight}⁻¹. In ¹³C-DNA, 531 Methylobacterium and Candidatus Methylopumilus were also detected, accounting for 532 0.01%-0.46% of the total sequencing reads. The relative abundance of methylotrophs 533 was high (28.6%) in the ¹³C-DNA of the M group at the CH₄ consumption of 20 μ mol 534 g_{wet weight}⁻¹, indicating that methylotrophs were the main microorganisms assimilating 535 CH₄-derived carbon. With increased incubation time, the relative abundance of 536 methylotrophs decreased in the ¹³C-DNA of the M group, indicating that more CH₄-537 derived carbon flowed into the non-methylotrophs in the SIP microcosms. In the MS 538 group, the relative abundance of methylotrophs increased and reached the maximum of 539

540 10.3% in the ¹³C-DNA at the CH₄ consumption of 50 μ mol g_{wet weight}⁻¹, which was lower 541 than that in the M group. This suggested that more CH₄-derived carbon flowed into the 542 non-methylotrophs such as CS₂-metabolizing microorganisms during CH₄ oxidation 543 with CS₂ existence.

The relative abundance of sulfur oxidizers including *Thiobacillus*, *Sulfuritalea*, 544 Sulfurifustis, Sulfuricella, Sulfuricurvum and Sulfuritortus was low (0.24%) in the 545 546 original sediment samples. After exposure to CS₂, the relative abundance of sulfur oxidizers increased to 1.8%-2.3% in the total DNA of sediment at the CH₄ consumption 547 of 50 µmol gwet weight⁻¹ and decreased with CH₄ consumption in the MS group (Fig. 8C). 548 Compared with the M group, the relative abundance of sulfur oxidizers was higher in 549 the MS group, indicating that more sulfur oxidizers took part in sulfur metabolism with 550 CS₂ existence. In the ¹³C-DNA, with the increasing CH₄ consumption, the relative 551 abundance of sulfur oxidizers increased from 0.75% to 2.14% when the CH₄ 552 consumption increased from 20 to 100 µmol gwet weight⁻¹ in the MS group. *Thiobacillus* 553 was the most abundant sulfur oxidizer in the ¹³C-DNA of the MS group at the CH₄ 554 consumption of 100 µmol g_{wet weight}⁻¹, suggesting that CH₄-derived carbon such as CO₂ 555 might serve as important carbon sources for Thiobacillus growth. Kappler and 556 Nouwens (2013) also found that facultative sulfur oxidizing chemolithoautotroph such 557 as Starkeya novella (formerly Thiobacillus novellus) could utilize C1 compounds 558 including methanol for growth and metabolism as well as sugars, amino sugars, amino 559 acids, and organic acids. Further studies need to be conducted to illustrate what are the 560 CH₄-derived carbon sources for *Thiobacillus*. 561

The diversity of sulfate-reducing bacteria was high but with a low proportion (Fig. 8D). *Pseudomonas*, *Geothermobacter*, *Desulfatiglans*, *Desulfobacca* and *Desulfobulbus* were the most abundant sulfate-reducing bacteria, accounting for 6.1%

of the relative abundance in the original sediment sample. After exposure to CH₄, the 565 relative abundance of sulfate-reducing bacteria decreased to 3.7% in the total DNA of 566 the M group, but increased to 6.5% in the total DNA of the MS group, when the CH₄ 567 consumption was 20 µmol gwet weight⁻¹. The relative abundances of sulfate-reducing 568 bacteria in the total DNA and ¹³C-DNA decreased with CH₄ consumption, which might 569 be attributed to aerobic condition inhibiting the growth of sulfate-reducing bacteria (Xu 570 571 et al., 2017). Compared with sulfur oxidizers, the relative abundance of sulfate-reducing bacteria was lower in the ¹³C-DNA of the MS group, indicating that sulfur oxidation 572 573 dominated in the SIP microcosm, which was in accordance with the high sulfate content of sediment under the existence of CS₂. 574

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576 4. Conclusion

H₂S and CS₂ were the two detected VSCs accompanied with CH₄ emitted from the 577 Meiliang Bay. SIP microcosm incubation showed although a CS₂ concentration of 350 578 mg m⁻³ did not have significant effect on the MOP, the fate of CH₄-derived carbon and 579 active microbial community varied much. Sediment CE_{CO2-C} was increased but 580 sediment CCE was decreased with the coexistence of CH₄ and CS₂. CS₂ was mainly 581 oxidized into SO₄²-S under aerobic condition. CS₂ could significantly change the 582 active microbial community structure involved into CH4-dervied carbon in the 583 sediments, with TS, SO42- and AVS levels serving as the main factors. CH4-derived 584 carbon mainly flowed into methylotrophs in the first stage. With CH₄ consumption, 585 more CH₄-derived carbon flowed into non-methylotrophs. CS₂ could prompt more 586 CH₄-derived carbon flowing into non-methanotrophs and non-methylotrophs, such as 587 sulfur-metabolizing bacteria. These findings can help elucidate the influence of VSCs 588 on microorganisms in fresh lake sediment and the responses of methanotrophs to 589

environmental changes, and provide insights to reducing CH₄ emission from eutrophiclake systems.

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593 **Declaration of competing interest**

594 The authors declare that they have no known competing financial interests or 595 personal relationships that could have appeared to influence the work reported in this 596 paper.

597

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