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Electronic Band Offset Determination of Oxides Grown by Atomic Layer Deposition on Silicon

Edris Khorani, Christoph A. Messmer, Sophie L. Pain, Tim Niewelt, Brendan F. M. Healy, Ailish Wratten, Marc Walker, Nicholas E. Grant, and John D. Murphy

Abstract-Minimizing electrical losses at metal/silicon interfaces in high efficiency single-junction silicon solar cells requires the use of carrier-selective passivating contacts. The electronic barrier heights at the insulator/silicon interface are necessary for calculating the probability of quantum tunnelling of charge carriers at these interfaces. Thus, precise knowledge of these parameters is crucial for the development of contact schemes. Using a photoemission-based method, we experimentally determine the electronic band offsets of Al₂O₃, HfO₂ and SiO₂ layers grown by atomic layer deposition (ALD) on silicon. For Al₂O₃/Si, we determine a valence band offset (ΔE_V) and conduction band offset (ΔE_C) of 3.29 ± 0.07 eV and 2.24 ± 0.13 eV, respectively. For HfO₂/Si, ΔE_V and ΔE_C are determined as 2.67 ± 0.07 eV and 1.81 \pm 0.21 eV, whilst for SiO₂/Si, ΔE_V and ΔE_C are 4.87 \pm 0.07 eV and 2.61 ± 0.12 eV, respectively. Using technology computer-aided design (TCAD) simulations, we incorporate our experimental results to estimate the contact resistivity that would be attained at various dielectric layer thicknesses. We find that for achieving the 100 mΩ.cm² contact resistivity benchmark, Al₂O₃ layers should be no thicker than 1.65 nm for a p-type polysilicon-based holeselective contact, assuming hole tunnelling masses taken from the literature. Correspondingly, for HfO2 and SiO2, an upper limit of 1.4 nm is determined as the thickness threshold in order to utilize these ALD-grown layers for contacts in high-performance silicon photovoltaics.

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I. INTRODUCTION

CCOUNTING for over 95% of commercial production, crystalline silicon solar cells continue to lead in the photovoltaics (PV) landscape and remain as a prominent alternative to non-renewable energy sources [1]. Currently, the bulk of this market is led by architectures that incorporate metal/Si interfaces for electrode formation [2], [3]. Avoiding this type of interface is widely accepted as the strategy towards higher performance silicon solar cells.

Due to the high density of electronically active states that arise from direct contact of metals on silicon, photogenerated charge carriers undergo trap-assisted recombination at such interfaces which limits the electrical device performance. To reach the Shockley-Queisser power conversion efficiency (PCE) limit of 29.4%, passivating and selective contact technologies are being adopted in the silicon PV industry [3]-[5]. Generally, passivating contacts are formed by introducing an interlayer or layer stack in between the silicon surface and the metal electrode. This type of technology mitigates the inherent electrical losses in the contacted regions by suppressing charge-carrier recombination as well as maintaining a low enough resistivity [6]-[8]. The use of a welldesigned passivating contact should increase the carrier collection efficiency and ultimately the PCE. Passivating both electron and hole contacts is needed in order to reach PCEs exceeding 25% [3], [9]. To date, various passivating contact structures have been successfully utilized in solar cell structures, including silicon heterojunctions (HJT) [10], polysilicon on oxide (POLO) [11] and tunnel oxide passivated contact (TOPCon) [12], [13]. Amongst these contact architectures, a PCE of 26.81% is the best cell performance achieved so far (using HJT) [14], [15].

Typically, SiO₂ based poly-Si contacts perform better as an electron-selective contact than a hole-selective contact. The valence band offset at the SiO₂/Si interface is considerably large, resulting in a high barrier for hole tunnelling and hence a relatively low hole tunnelling current [16]. Another reason for the poorer performance as a hole-selective contact is due to the high temperature anneal step which causes dopants to diffuse from the poly-Si layer to the Si substrate. As boron diffusion in SiO₂ is not blocked as well as phosphorous, a relatively large boron concentration is found at the SiO₂/Si interface in this contact architecture which in-turn leads to Auger recombination [17]. Despite the key improvements over recent years in

passivating contact technologies, the search for a hole-selective contact that can match or even exceed the performance of existing electron-selective contacts continues.

In thin film passivating contacts, the exact mechanism for charge carrier transport is still under debate in the PV community, but generally it is understood to be due to quantum tunnelling and/or through pinholes [18]–[20]. Tunnelling refers to the ability of charge carriers to have a wavefunction that can extend through a potential barrier instead of the carrier having to go over the barrier. The probability of carrier tunnelling, P_{t} , through such insulators is described by the Wentzel-Kramers-Brillouin (WKB) approximation [21]:

$$P_t = \exp\left(-\frac{2}{\hbar}t\sqrt{2m^*q\Delta\phi_b}\right) \tag{1}$$

where \hbar is the reduced Planck's constant, m^* is the tunnelling charge effective mass, q is the charge of an electron, t is the film thickness and $\Delta \phi_b$ is potential barrier height at the interface. From the WKB approximation, it follows that the tunnelling probability can be controlled by tuning the film thickness and potential barrier heights at the insulator/Si interface.

Utilizing thin films for carrier-selective passivating contacts requires fabrication techniques capable of attaining thicknesses with Angstrom (Å) level controllability in a reliable manner. Atomic layer deposition (ALD) offers such benefits through self-limiting surface reactions conducted at relatively low deposition temperatures. It is highly suitable for depositing a variety of materials, including oxides and nitrides, which fit the criteria for contact interlayers. Ultimately, the high level of film and interface control is highly attractive for thin film fabrication for PV applications, particularly for carrier-selective passivating contacts.

In previous work, we report the Si surface passivation quality of various types of ultrathin dielectrics grown via ALD [22]. HfO₂ was identified as the most promising candidate, with 0.9 nm of HfO₂ annealed at 450 °C providing a surface recombination velocity (SRV) of 18.6 cm s⁻¹ and 2.2-3.3 nm thick HfO₂ layers achieving an SRV of ≤ 2.5 cm s⁻¹ and J₀ of ~ 14 fA/cm². As the development of an efficient hole-selective contact is a key aim in today's PV industry, we determine the potential for these materials as carrier-selective interlayers. In this paper, we report an experimental study that determines the electronic barrier heights at the interface between silicon and ALD-grown Al₂O₃, HfO₂ and SiO₂. We use our existing ALD growth methods [22], [23] to grow thin films on silicon. We then use X-ray photoelectron spectroscopy (XPS) to probe the electronic core levels (CL) and valence band maximum (VBM) at the oxide/Si interface. Additionally, we use simulations in Sentaurus TCAD [24] to estimate the contact resistivity that we would expect in a Si/oxide/poly-Si p-contact formation for each of the investigated oxides. Since the barrier tunnelling model in Sentaurus is based on the WKB approximation (1), the resulting contact resistivities highly depend on the oxide thickness, t, as well as the effective (hole) tunnelling mass, $m_{\rm h}$ *. Assuming hole tunneling masses from literature, these simulations indicate that reasonable contact properties could be achieved and therefore provide an incentive for further experimental

research on the development of an efficient hole-selective contact based on these oxides. Beyond minimizing contact resistivity, the use of these dielectrics as interlayers in carrierselective passivating contacts requires the enhancement of the surface passivation properties of these films at ultra-thin (sub-3 nm) thicknesses, as explored in Ref. [22].

II. EXPERIMENTAL METHODS

A. Specimen Fabrication

ALD films were grown on p-type (gallium doped) Si (Cz, 5 Ω cm, <100>, 125 μ m thick) substrates that were prepared following a previously reported chemical cleaning and etching procedure [22], [25]. In the first ALD half-cycle reaction, trimethylaluminum, tetrakis(dimethylamido)hafnium and bis(diethylamido)silane precursors were used to grow Al₂O₃, HfO₂ and SiO₂ films, respectively. An O₂ plasma source was used for the second half-cycle reaction for all three films. All films were grown at a deposition temperature of 200 °C. Growth rates per cycle are reported as 1.3 Å/cycle (Al_2O_3), 1 Å/cycle (HfO₂) and 0.6 Å/cycle (SiO₂) [26], and re-evaluated in [22]. A post-deposition anneal was conducted at 450 °C for 30 minutes for Al₂O₃ and HfO₂, and 800 °C for 30 minutes for SiO2. All films were grown in a Veeco Fiji G2 plasma-enhanced ALD chamber.

B. X-ray Photoelectron Spectroscopy

XPS was conducted using a Kratos Axis Ultra DLD spectrometer. For XPS, all samples were mounted on a nonmagnetic, stainless-steel bar by using electrically conductive carbon tape. XPS was conducted using a monochromated Al Ka X-ray (1.487 keV) source. The energy resolution of the detector was 0.4 eV. Measurements were conducted at room temperature and at a take-off angle of 90° with respect to the sample surface. The CL spectra and the VBMs were measured using a pass energy of 20 eV, all from an analysis area of 300 μ m \times 700 μ m. To avoid charging effects, a charge neutraliser gun was used for all XPS measurements. Fitting procedures to extract peak positions and relative stoichiometries were performed by using the Casa XPS software. These were fitted and corrected using their corresponding sensitivity factors, taking the mean free path of the photoelectrons and photoionization cross sections of these core levels into account.

C. Electronic Band Offsets Determination

Determination of the valence band offset (ΔE_V) and conduction band offset (ΔE_C) at a semiconductor interface can be done using Kraut's method [27]–[31]. These are depicted in the schematic diagram of the band offsets in Fig. 1.

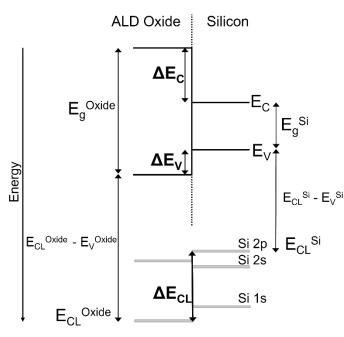


Fig. 1. Schematic diagram of band offsets at an oxide/Si interface.

This X-ray photoemission-based method uses Poisson's equation to predict the band discontinuities based on the deviations in charge distribution found at the interface relative to the semiconductor bulk. In this approach, the position of the core level at the interface, as well as the binding energy difference between the semiconductor CL and the VBM are required to determine the valence band offset:

$$\Delta E_{v} = \left(E_{CL}^{Si} - E_{CL}^{0xide}\right)_{0xide/Si} - \left(E_{CL}^{Si} - E_{V}^{Si}\right)_{Si} + \left(E_{CL}^{0xide} - E_{V}^{0xide}\right)_{0xide}$$
(2)

In this equation, $(E_{CL}^{Si} - E_{CL}^{Oxide})_{Oxide/Si}$ is the energy difference between the CL of the two materials at the interface, namely ΔE_{CL} . Based on the XPS photoelectron sampling depth being under 5 nm in these ALD-grown materials [32], [33], 2 nm and 3 nm thick films on Si were used to obtain two sets of measurements that probe the interface. To complete the equation for ΔE_V , the energy difference between the CL centroids and VBM for Si and all the ALD oxides of interest were obtained from XPS of the respective thick films. For this experiment, the Al 2p, Hf 4f and Si 2p orbital peaks were used as the CL for Al₂O₃, HfO₂ and SiO₂, respectively. The elemental silicon region of the Si 2p peak was also used as the CL for Si. For E_V^{Si} and E_V^{Oxide} determination, linear extrapolation of the leading edge to the baseline of the valence band spectra from the respective thick films were used.

Once ΔE_V was determined, ΔE_C was calculated following Kraut's method [27]:

$$\Delta E_C = \Delta E_V - \left(\Delta E_g\right)_{Si/Oxide} \tag{3}$$

where $(\Delta E_g)_{Si/Oxide}$ is the energy difference between the band gap of Si and the respective ALD oxide films. An optical band gap of 1.12 ± 0.01 eV was used for E_g^{Si} in these calculations.

III. RESULTS AND DISCUSSION

A. Electronic Band Offsets

The electronic band offsets at a Al_2O_3/Si , HfO_2/Si and SiO_2/Si interface have been reported in the literature. A summary of these calculations from previous reports is presented in Table I.

TABLE I REPORTED VALENCE BAND OFFSET (ΔE_V) AND CONDUCTION BAND OFFSET (ΔE_C) FOR AL₂O₃/SI, HFO₂/SI and SIO₂/SI TAKEN FROM LITERATURE.

| Al ₂ O ₃ /Si | | HfO ₂ /Si | | SiO ₂ /Si | | |
|------------------------------------|----------------|----------------------|----------------|----------------------|-------------------------|--|
| ΔEv | ΔE_{C} | ΔEv | ΔE_{C} | ΔEv | $\Delta E_{\rm C} (eV)$ | |
| (eV) | (eV) | (eV) | (eV) | (eV) | | |
| 4.9 [34] | 2.8 [34] | 3.4 [34] | 1.5 [34] | 4.54 [16] | 3.15 [16] | |
| 4.1 [35] | 3.5 [35] | 2.87 [36] | 1.71 [36] | 4.49 [37] | 3.29 [37] | |
| 3.5 [38] | - | 2.69 [39] | 2.0 [39] | 4.4 [34] | 3.5 [34] | |
| 3.24 [36] | 2.44 [36] | 2.5 [40] | 2.2 [40] | 4.3 [38] | - | |
| 2.95 [41] | 2.1 [41] | 2.5 [42] | 2.0 [42] | _ | 3.13 [43] | |
| - | 2.13 [44] | | | | | |

A large range of band offsets have been presented over the last couple of decades. For example, ΔE_V at a Al₂O₃/Si interface has been reported between 2.95 eV and 4.9 eV. The differences are mainly due to dissimilarities in fabrication processes that result in variations in chemical composition and stoichiometry that in turn lead to alterations in the band offsets. For example, Alay et al. suggest differences in the band offsets for SiO₂/Si based on whether the SiO₂ layer was fabricated by a dry or wet chemical process as well as the crystal orientation of the underlying Si substrate being (100) or (111) [37]. Interfacial effects such as interfacial dipoles could also lead to deviations in band offsets. For ΔE_{CL} determination (as part of (2)), the thickness of the overlayer chosen can also play a role in the band offset calculations. It is generally understood that the thickness must not exceed the photoelectron sampling depth of the overlaying material, but variations in thickness below that limit can cause small shifts in the CL positions. Also, differences in measurement procedures (e.g., XPS, linear internal photoemission, and synchrotron radiation photoemission) add further uncertainty in the reported band offsets. To accurately determine the probability of carrier tunnelling through such interfaces, precise determination of the band offsets is required.

Using XPS, we have identified the core level energy centroids and valence band edges for Si, Al₂O₃, HfO₂ and SiO₂, as shown in Fig. 2. From Fig. 2 (a) and 2 (b), the Si 2p CL energy and leading edge of the valence band spectra (i.e. VBM) for bare (native oxide stripped with HF) Si are determined to be 99.71 \pm 0.02 eV and 0.69 \pm 0.04 eV, respectively. The \pm symbol is used to signify the measurement uncertainty. Hence, (E_{CL}^{Si} – E_V^{Si})_{Si} is calculated to be 99.02 \pm 0.045 eV. The propagated uncertainty is determined as the square root of the sum of the

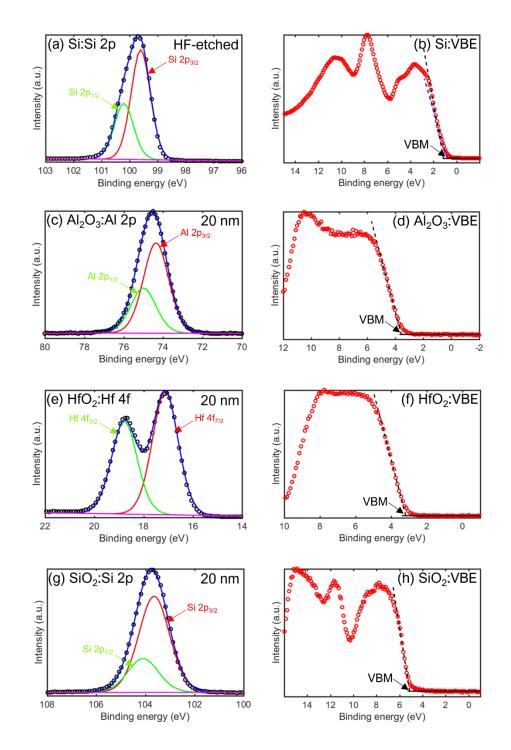


Fig. 2. XPS spectra showing: (a) Si 2p CL and (b) VBM for bulk *p*-type Si, (c) Al 2p CL and (d) VBM for bulk Al_2O_3 , (e) Hf 4f CL and (f) VBM from bulk HfO₂ and (g) Si 2p CL and (h) VBM for bulk SiO₂. Circle symbols denote measurements, solid lines denote fitted subcomponents and their superposition and dashed lines illustrate the extraction of the VBM.

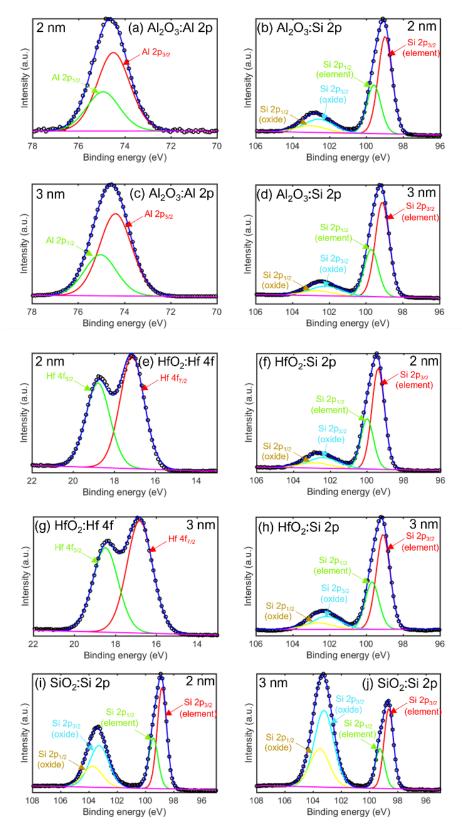


Fig. 3. XPS spectra showing (a) Al 2p CL and (b) Si 2p CL from 2 nm Al_2O_3 on Si, (c) Al 2p CL and (d) Si 2p CL from 3 nm Al_2O_3 on Si, (e) Hf 4f CL and (f) Si 2p CL from 2 nm HfO₂ on Si, (g) Hf 4f CL and (h) Si 2p CL from 3 nm HfO₂ on Si, (i) Si 2p CL from 2 nm SiO₂ on Si and (j) Si 2p CL from 3 nm SiO₂ on Si. Circle symbols denote measurements, solid lines denote fitted subcomponents and their superposition.

squares of E_{CL} and E_V . Fig. 2 (c) and 2 (d) show the Al 2p CL energy and VBM for Al₂O₃. The Al 2p CL is determined at a binding energy of 74.50 ± 0.02 eV and the VBM at 3.37 ± 0.04 eV. From this, $(E_{CL}^{Al2O3} - E_V^{Al2O3})_{Al2O3}$ is calculated to be 71.13 ± 0.045 eV. For bulk HfO₂, the Hf 4f CL energy and VBM, shown in Fig. 2 (e) and 2 (f), are determined as 17.12 ± 0.02 eV and 3.15 ± 0.04 eV, respectively. Hence, $(E_{CL}^{HfO2} - E_V^{HfO2})_{HfO2}$ is calculated to be 13.97 ± 0.045 eV. From Fig. 2 (g) and 2 (h), the Si 2p CL energy and VBM for SiO₂ are determined as 103.68 ± 0.02 eV and 5.02 ± 0.04 eV, respectively. Therefore, $(E_{CL}^{SiO2} - E_V^{SiO2})_{SiO2}$ is calculated to be 98.66 ± 0.045 eV. A summary of the XPS data taken from Fig. 2 can be found in Table II.

TABLE II PARAMETERS EXTRACTED FROM THE XPS DATA SHOWN IN FIG. 2 FOR ΔE_V CALCULATIONS.

| Bulk material | Ecl (eV) | Ev (eV) | (Ecl-Ev) (eV) |
|--------------------------------|------------------|-----------------|------------------|
| Silicon | 99.71 ± 0.02 | 0.69 ± 0.04 | 99.02 ± 0.045 |
| Al ₂ O ₃ | 74.50 ± 0.02 | 3.37 ± 0.04 | 71.13 ± 0.045 |
| HfO ₂ | 17.12 ± 0.02 | 3.15 ± 0.04 | 13.97 ± 0.045 |
| SiO ₂ | 103.68 ± 0.02 | 5.02 ± 0.04 | 98.66 ± 0.045 |

The CL for Al₂O₃ (Al 2p) and Si (Si 2p) at the interface from the 2 nm and 3 nm thick Al₂O₃ specimens are shown in Fig. 3 (a)-(d). From Fig. 3 (b) and 3 (d), two peaks are seen for the Si 2p CL at the interface. The CL found at ~99 eV is detected from elemental silicon and the CL at ~103 eV is from the presence of SiO₂. This CL verifies the presence of SiO₂ in these specimens, suggesting the manifestation of a very thin SiO₂ layer at the Al₂O₃/Si interface [45]. For this study, we focus on only using the elemental silicon region for E_{CL}^{Si} in the band offset calculations.

From Fig. 3 (a) and 3 (c), the Al 2p CL from 2 nm and 3 nm thick Al₂O₃ on Si are found at binding energies of 74.62 \pm 0.02 eV and 74.55 \pm 0.02 eV, respectively. Additionally, Fig. 3 (b) and 3 (d) show the Si 2p CL, with the binding energies found at 99.12 \pm 0.02 eV and 99.25 \pm 0.02 eV. Hence, (E_{CL}^{Si} – E_{CL}^{Al2O3})_{Al2O3/Si} is determined for the 2 nm and 3 nm Al₂O₃ specimen as 24.50 \pm 0.028 eV and 24.70 \pm 0.028 eV, respectively.

Fig. 3 (e)-(h) show the Hf 4f and Si 2p CL from 2 nm and 3 nm thick HfO₂ on Si. From Fig. 3 (e) and 3 (g), the Hf 4f CL binding energies are determined as 17.13 ± 0.02 eV and 16.81 ± 0.02 eV. The Si 2p CL in Fig. 3 (f) and 3 (h) show a peak at ~103 eV as well as the Si 2p elemental silicon peak at ~99 eV, which again demonstrates the presence of an interfacial SiO₂ layer. Taking the elemental silicon contribution into account, the Si 2p CL for 2 nm and 3 nm thick HfO₂ on Si are found at 99.46 \pm 0.02 eV and 99.23 \pm 0.02 eV, respectively. Therefore, (E_{CL}^{Si} – E_{CL}^{HfO2})_{HfO2/Si} for 2 nm and 3 nm thick HfO₂ on Si are determined to be 82.33 \pm 0.028 eV and 82.42 \pm 0.028 eV, respectively.

Fig. 3 (i) and 3 (j) show the Si 2p CL peaks from 2 nm and 3 nm thick SiO_2 on Si, respectively. Here, we only take the Si

2p into account as we detect contributions from elemental Si and Si-O in the same binding energy region. From Fig. 3 (i) and 3 (j), the Si 2p (Si-O) CL peaks are found at binding energies of 103.28 \pm 0.02 eV and 103.23 \pm 0.02 eV. Si 2p (elemental) peaks are detected at binding energies of 98.83 \pm 0.02 eV and 98.65 \pm 0.02 eV. Hence, for 2 nm and 3 nm thick SiO₂ on Si, $(E_{CL}^{Si} - E_{CL}^{SiO2})_{SiO2}$ are calculated to be -4.45 \pm 0.028 eV and -4.58 \pm 0.028 eV, respectively. A summary of the XPS data and corresponding calculations taken from Fig. 3 are shown in Table III.

TABLE III XPS DATA TAKEN FROM FIG. 2 AND 3 FOR ΔE_V CALCULATIONS

| Interface | Ecl ^{Si} | EcL ^{Oxide} | ECL ^{Si} - ECL ^{Oxide} |
|---|-------------------|----------------------|--|
| Interface | ECL" | ECL | ECL - ECL |
| (2nm)Al ₂ O ₃ /Si | 99.12 ± 0.02 | 74.62 ± 0.02 | 24.50 ± 0.028 |
| (3nm)Al ₂ O ₃ /Si | 99.25 ± 0.02 | 74.55 ± 0.02 | 24.70 ± 0.028 |
| (2 nm)HfO ₂ /Si | 99.46 ± 0.02 | 17.13 ± 0.02 | 82.33 ± 0.028 |
| (3 nm)HfO ₂ /Si | 99.23 ± 0.02 | 16.81 ± 0.02 | 82.42 ± 0.028 |
| (2 nm)SiO ₂ /Si | 98.83 ± 0.02 | 103.28 ± 0.02 | -4.45 ± 0.028 |
| (3 nm)SiO ₂ /Si | 98.65 ± 0.02 | 103.23 ± 0.02 | -4.58 ± 0.028 |

From Tables II and III, ΔE_V and ΔE_C are calculated using (2) and (3) respectively and are shown in Table IV. To determine ΔE_C , we use a bandgap of 6.65 \pm 0.11 eV for Al₂O₃ [46], 5.6 \pm 0.2 eV for HfO₂ [47], [48] and 8.6 \pm 0.1 eV for SiO₂ [49]. For (E_{CL}^{Si} - E_{CL}^{Oxide})_{Oxide/Si} in (2), an average is taken between the 2 nm and 3 nm oxide/Si calculations from Table III. From these results, Fig. 4 shows a simplified schematic diagram of the band offsets at the Al₂O₃/Si, HfO₂/Si and SiO₂/Si interface.

 TABLE IV

 CALCULATED VALENCE AND CONDUCTION BAND OFFSETS.

| CALCOLATED VALLENCE AND CONDUCTION DATED OF SETS | | | | |
|--|------------------|-------------------------|--|--|
| Interface | $\Delta Ev (eV)$ | $\Delta E_{\rm C} (eV)$ | | |
| Al ₂ O ₃ /Si | 3.29 ± 0.07 | 2.24 ± 0.13 | | |
| HfO ₂ /Si | 2.67 ± 0.07 | 1.81 ± 0.21 | | |
| SiO ₂ /Si | 4.87 ± 0.07 | 2.61 ± 0.12 | | |

The $\Delta E_C / \Delta E_V$ ratio is a good indication of favorability towards electron/hole transport, where $\Delta E_C / \Delta E_V > 1$ favors hole transport. For Al₂O₃/Si, HfO₂/Si and SiO₂/Si, $\Delta E_C/\Delta E_V$ can be determined as 0.68, 0.68 and 0.54, respectively. This suggests that all three interfaces would favor electron transport, with SiO_2 being the most favorable towards electrons. SiO_2 is used as the passivating interlayer in the world-leading carrierselective passivating contact technology, TOPCon, and is known to perform far better as an electron contact than as a hole contact [3]. The band offsets determined indicate that HfO₂ and Al₂O₃ could offer alternatives for SiO₂ in the hole-selective counterpart. Beyond $\Delta E_{\rm C}/\Delta E_{\rm V}$, a direct comparison of the absolute values for ΔE_V determined for the three oxides of interest is another good indication towards hole transport favorability (based on (1)). Evidently, HfO₂ possesses the smallest potential barrier for hole transport but a ΔE_V of 2.67 eV is still considerably large.

The band offsets are not the only figure of merit when considering fitting candidates for carrier-selective contacts so

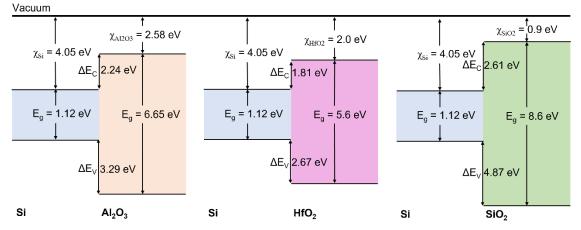


Fig. 4. Simplified schematic diagram of measured conduction band offset and valence band offsets of ALD-deposited Al₂O₃, HfO₂ and SiO₂ on Si.

the impact of this must be weighed with their suitability in other important factors, including carrier tunnelling mass and surface recombination velocity. We explored the Si surface passivation quality in previous work [22] and determine the impact of the band offsets and tunnelling masses on the contact resistivity using TCAD simulations here.

B. Contact Resistivity Estimation via TCAD Simulations

For charge carrier transport through ultrathin dielectrics, one of the theories that is strongly agreed upon is quantum tunnelling through the potential barrier created at the Si surface. Based on the WKB approximation (1), the tunnelling probability is dependent on the barrier height and film thickness, as well as the effective tunnelling masses of the charge carriers.

Using the barrier heights determined for the ALD oxides with respect to Si, Sentaurus TCAD is used to estimate the contact resistivity for a range of thicknesses in a typical p-TOPCon format. The contact resistivity is extracted directly from the TCAD simulations for this study, An illustration of the contact structure (i.e. poly-Si/oxide/Si) we are interested in is shown in Fig. 5 (a). A *p*-type Si (1 Ω cm, 200 μ m thick) substrate with the ALD oxides on the front side was devised, with a 50 nm thick p^+ poly-Si conductive interlayer (with 10^{20} cm⁻³ doping concentration) between the metal contact and the oxide. In the TCAD simulations, an electron effective mass of 0.25 m₀ [46], a hole effective mass of 0.36 m₀ [50] and an optical bandgap of 6.65 eV [46] were used for Al_2O_3 , where m_0 is the free electron mass. For HfO2, an electron effective mass of 0.11 m₀ [51], a hole effective mass of 0.58 m₀ [51], [52] and an optical bandgap of 5.6 eV [47], [48] were used as taken from literature. For SiO₂, an electron effective mass of 0.4 m_0 [53], [54], a hole effective mass of 0.3 m_0 [55] and an optical bandgap of 8.6 eV were used. Fig. 5(b) shows the tunnelling layer thickness vs. calculated contact resistivity (ρ_c) for ALD-grown Al₂O₃, HfO₂, and SiO₂ on p-Si as a hole-selective contact.

The calculated ρ_c vs dielectric thickness curves show an exponential trend in all three cases. The difference in ρ_c between the three ALD-grown dielectrics below 1.1 nm is negligible. From Fig. 5(b), the Al₂O₃/(p)Si based contact

outperforms HfO₂ and SiO₂ (i.e., the lowest ρ_c) at most thicknesses. Despite the larger ΔE_V measured for Al₂O₃/Si in comparison to HfO₂/Si, the Al₂O₃/Si contact outperforms HfO₂/Si due to the considerably lower hole effective mass. This demonstrates how crucially the contact resistivity depends on the assumed tunnelling masses. Based on (1), the tunnelling effective mass and interfacial barrier heights are of equal importance and ideally, both parameters should be low enough to ensure a low contact resistivity.

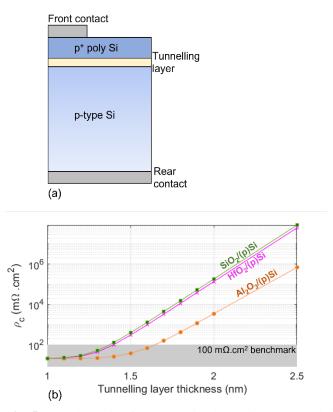


Fig. 5. (a) Schematic of structure simulated via TCAD and (b) tunnelling layer thickness vs contact resistivity for ALD-grown Al_2O_3 , HfO₂ and SiO₂ on *p*-type Si.

As a standard for carrier-selective passivating contacts, a contact resistivity of 100 m Ω .cm² is seen as the upper threshold

for high-performance full-area contacts. Achieving a contact resistivity below 100 m Ω .cm² provides insignificant improvements to the fill factor, and hence the PCE of such devices. Based on our TCAD simulations with the given effective tunnelling masses from literature, a thickness of 1.65 nm is found as the upper limit for Al₂O₃. The upper limit for both HfO₂ and SiO₂ is determined as 1.4 nm.

Incorporating metal oxides like Al₂O₃ with poly-Si for holeselective passivating contacts have seen some success in literature, with the contact resistivity reported at 200 m Ω .cm² in multiple findings in literature [56]-[58]. This further illuminates the scope for research on Al2O3-based holeselective passivating contacts. For HfO₂, despite showing a higher contact resistivity than Al₂O₃ in our TCAD simulations, the surface recombination velocities determined for ultrathin HfO₂ layers outperform Al₂O₃, operating just as well as thicker HfO₂ films [59], as found in our previous work [22], [60]. These results were achieved without common post-deposition treatments like hydrogenation, which are typically used to enhance the surface passivation properties [22], [57], [61]. However, the impact of growing a p⁺ poly-Si layer on the oxides as well as annealing for crystallizing the poly-Si layer at temperatures exceeding 800 °C is yet to be examined. Also, the use of ALD-grown SiO₂ as a replacement for thermally grown SiO₂ still requires substantial improvements, mainly due to the inferior passivation quality of this type of SiO₂ growth method [13], [22], [57].

IV. CONCLUSION

We explored the electronic band offsets of ALD-grown Al₂O₃, HfO₂ and SiO₂ on silicon using a photoemission-based method. For Al₂O₃/Si, we determine ΔE_V and ΔE_C as 3.29 ± 0.07 eV and 2.24 \pm 0.13 eV, respectively. For HfO₂/Si, ΔE_V and ΔE_{C} are determined as 2.67 \pm 0.07 eV and 1.81 \pm 0.21 eV, whilst for SiO₂/Si ΔE_V and ΔE_C are 4.87 ± 0.07 eV and 2.61 ± 0.12 eV, respectively. We apply TCAD simulations to predict the contact resistivity at various dielectric thicknesses with a 50 nm p⁺ polycrystalline silicon conductive layer between the metal electrode and thin dielectrics and assuming hole effective tunnelling masses taken from literature. In order to form an efficient hole-selective contact with a p-type polysilicon electrode and to not exceed the 100 m Ω .cm² contact resistivity benchmark, an upper limit of 1.65 nm in thickness is determined for Al₂O₃, whilst 1.4 nm is calculated as the threshold for HfO₂ and SiO₂. Experimental demonstration of the contact resistivity for such structures will be part of future work.

DATA ACCESS STATEMENT

Data underpinning figures in this paper can be downloaded from <u>https://wrap.warwick.ac.uk/75494</u>. Requests for additional data should be made directly to the corresponding author. For the purpose of open access, the author has applied a Creative Commons Attribution (CC BY) licence to any Author Accepted Manuscript version arising from this submission. REFERENCES

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